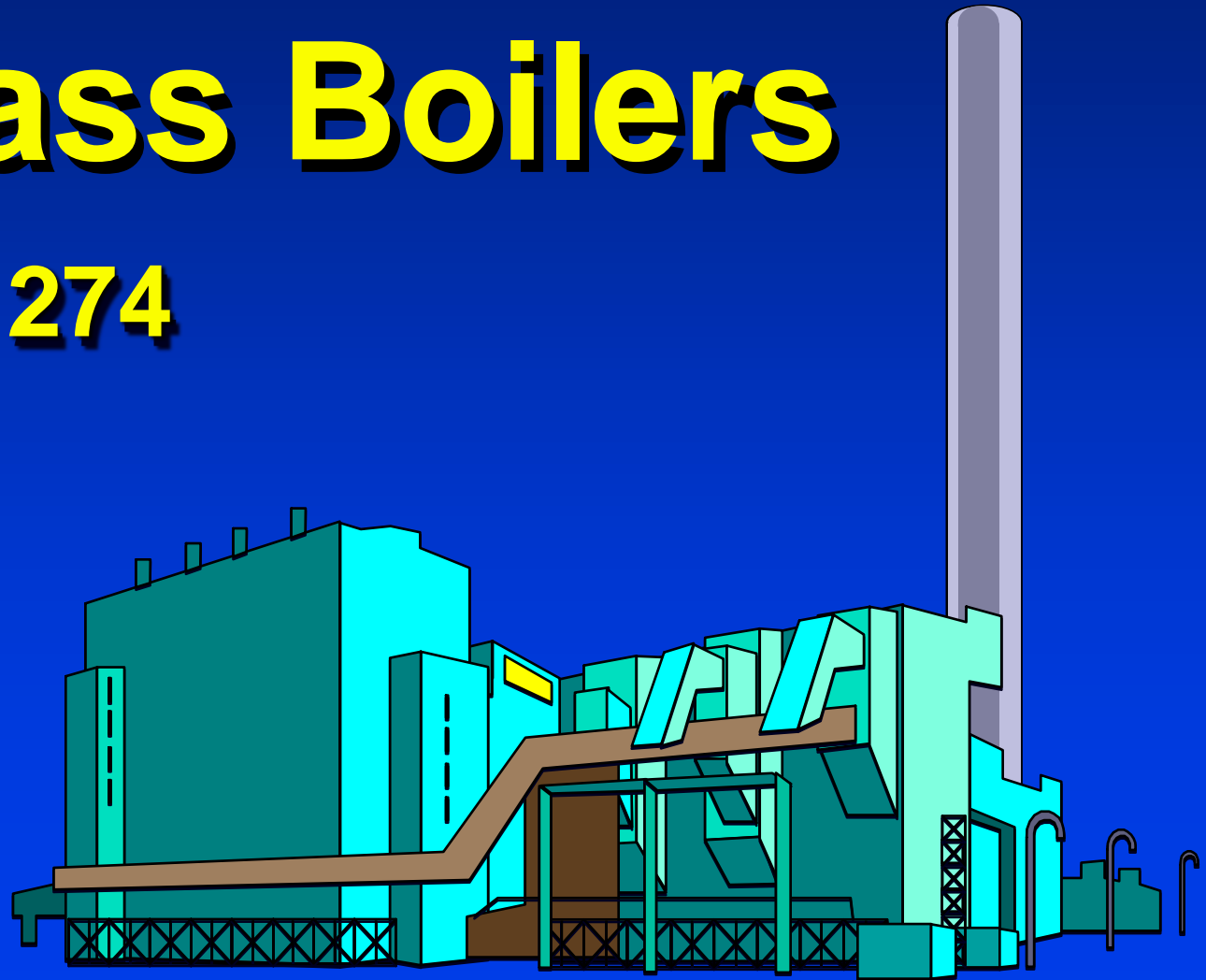




California Environmental Protection Agency

# Biomass Boilers

Course # 274

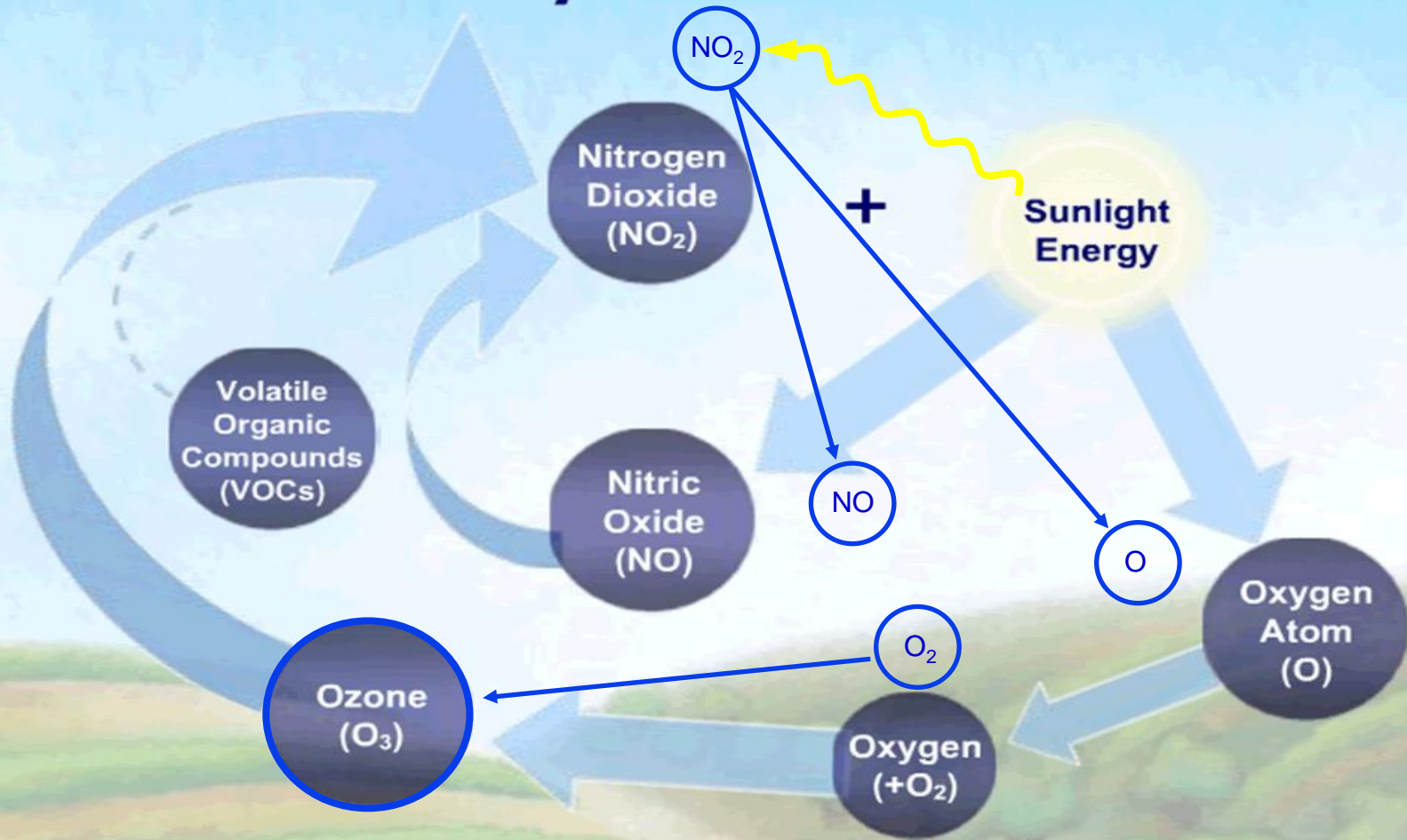


# Course Overview

- ◆ Air Pollution *Why*
- ◆ Boiler Uses *What*
- ◆ Boiler Theory and Operation
- ◆ Air Pollution Formation
- ◆ Air Pollution Control Devices
- ◆ Boiler Regulations *How*
- ◆ Typical Permit Conditions
- ◆ Inspection Procedures



# Ozone Photochemistry



# Uses of Boilers

- ◆ **Electrical generation** High Pressure (1,800 -3,800 psi)
- ◆ **Space heating** Low Pressure (150 – 1,600 psi)
- ◆ **Food preparation**
- ◆ **Commercial laundries**
- ◆ **Pulp & paper industry**
- ◆ **Petroleum industry**
- ◆ **Chemical industry**



**Small  
Firetube  
Boiler**

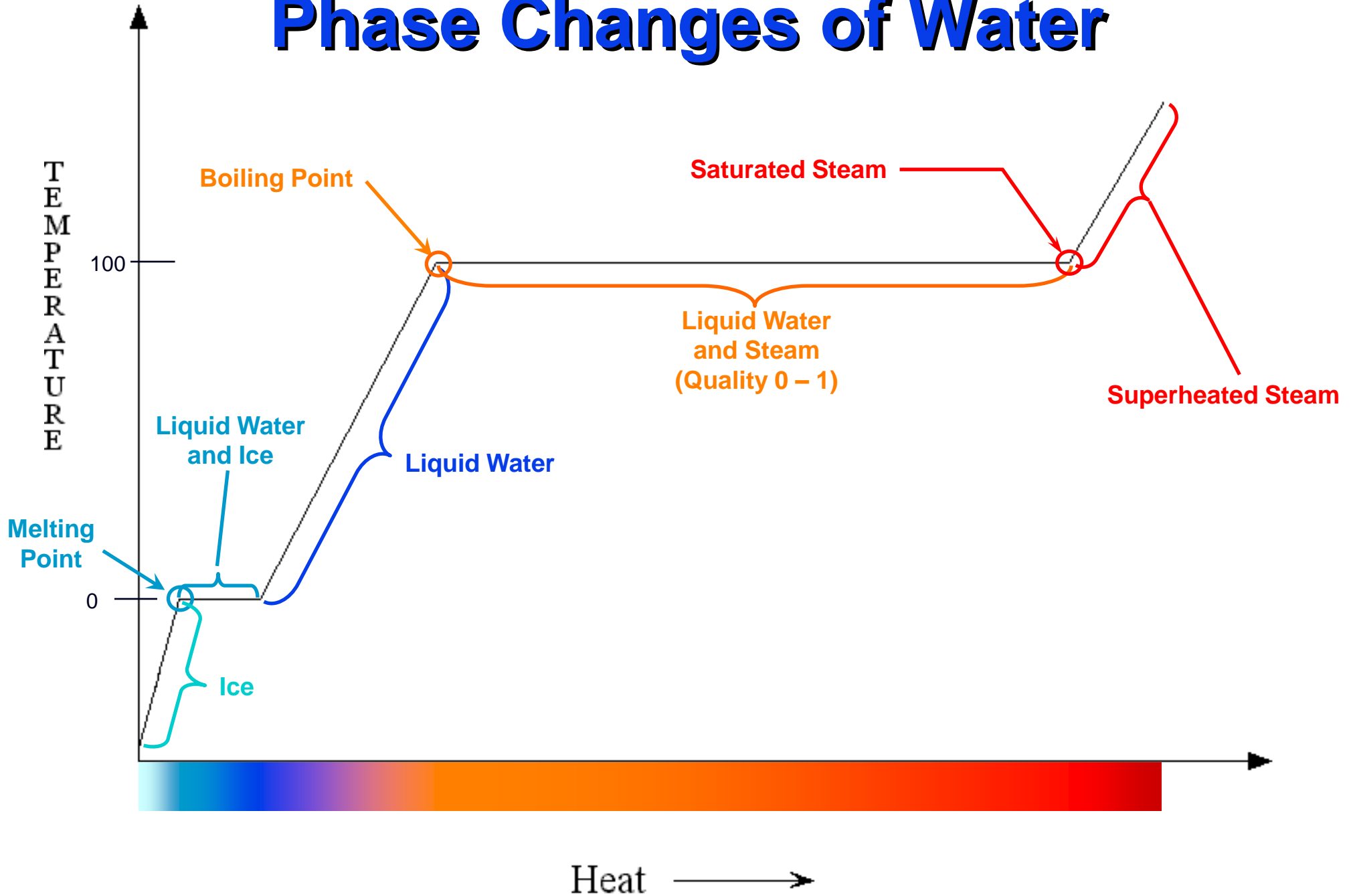


# Industrial Boiler



**Biomass  
Utility  
Boiler**

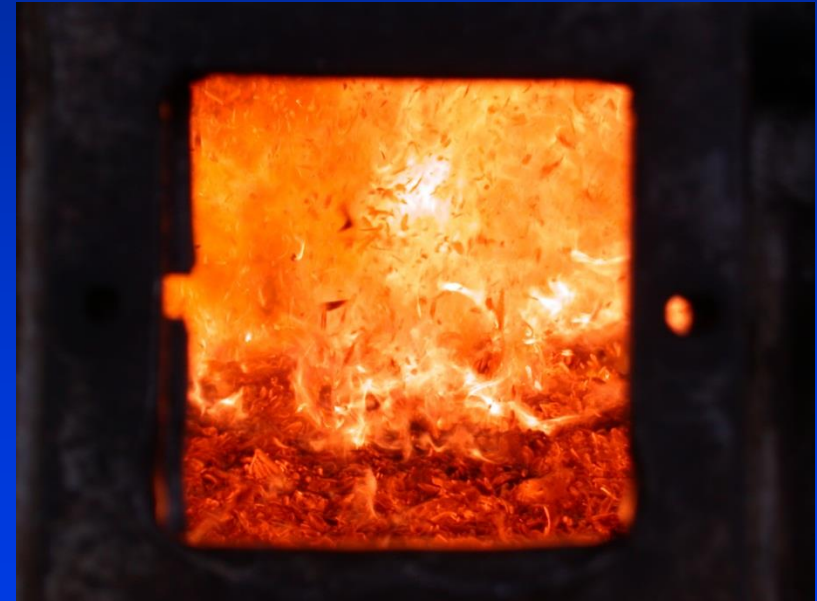
# Phase Changes of Water

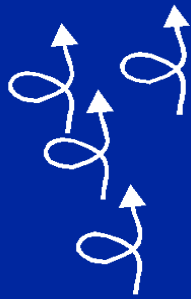




# Hot Numbers

- ◆ **British Thermal Unit (BTU)**
  - ◆ 1 BTU the amount of energy needed to heat one pound of water one degree Fahrenheit or  $\approx$  energy given off by burning one wooden match
- ◆ **Lower Heating Value (LHV)**
  - ◆ Heating value of a fuel not counting heat needed to vaporize water
- ◆ **Higher Heating Value (HHV)**
  - ◆ Heating value of a fuel including heat needed to vaporize water



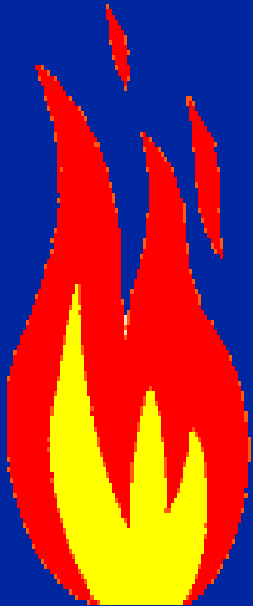


**CONVECTION**  
HOT GASES  
TRANSFER HEAT  
TO THE TUBE

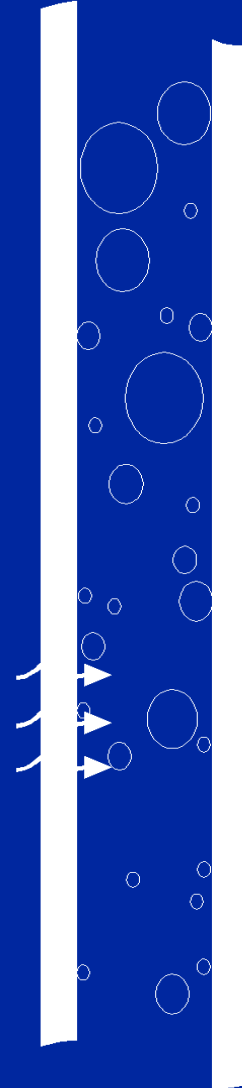
**RADIATION**



HEAT TRANSFER  
THRU SPACE



**CONDUCTION**  
HEAT TRANSFER  
THRU THE  
METAL TUBE  
WALL



# Heat Transfer Methods



**Boilers &  
Opacity**

# What Opacity is This?





# **Boiler Ratings**

- Millions of BTU/hr**
- Boiler HP**
- Pounds of Steam/hr**
- Megawatts**
- Tons per day**


# COMBUSTION ENGINEERING, INC.

NEW

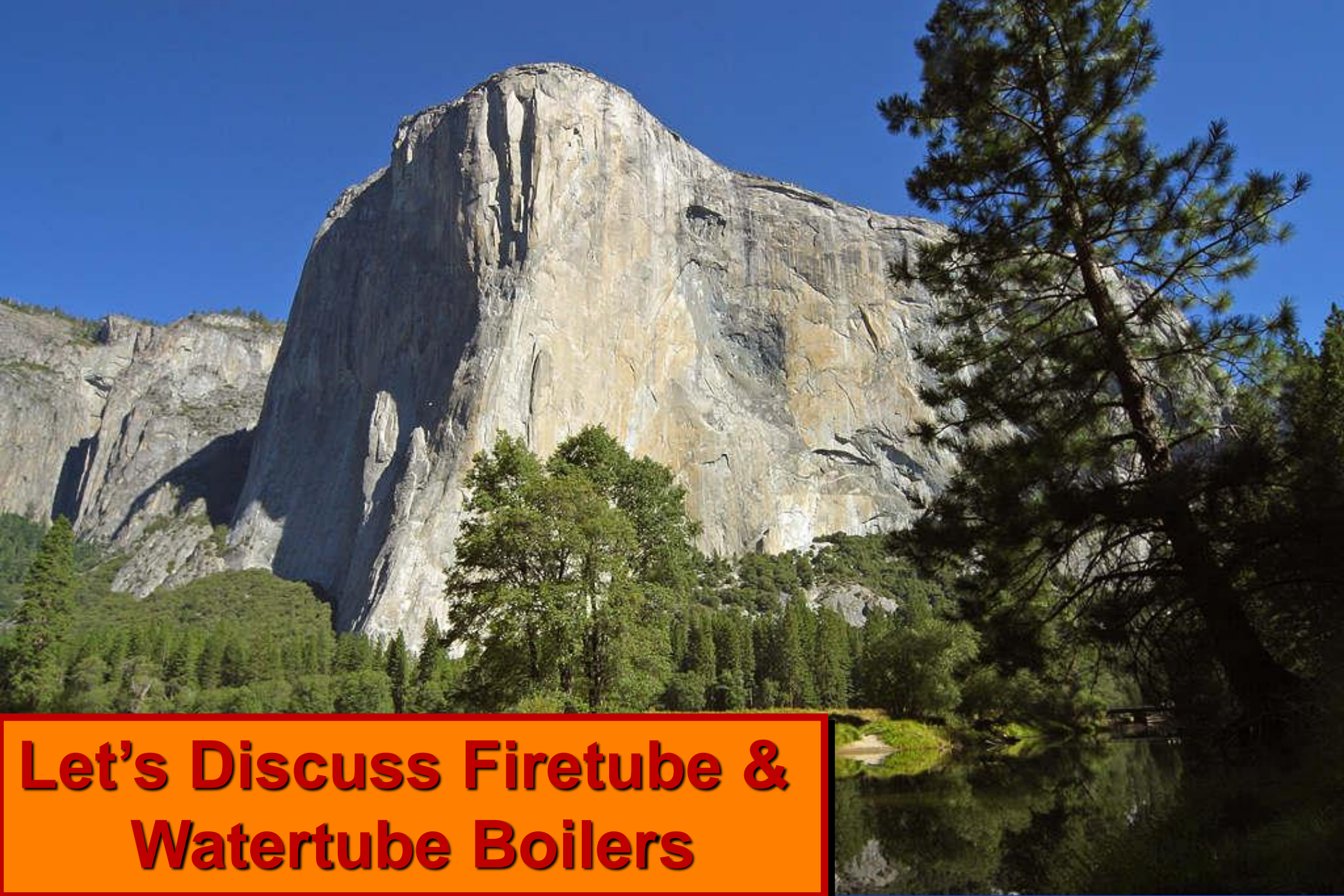


YORK

## C-E STEAM GENERATOR

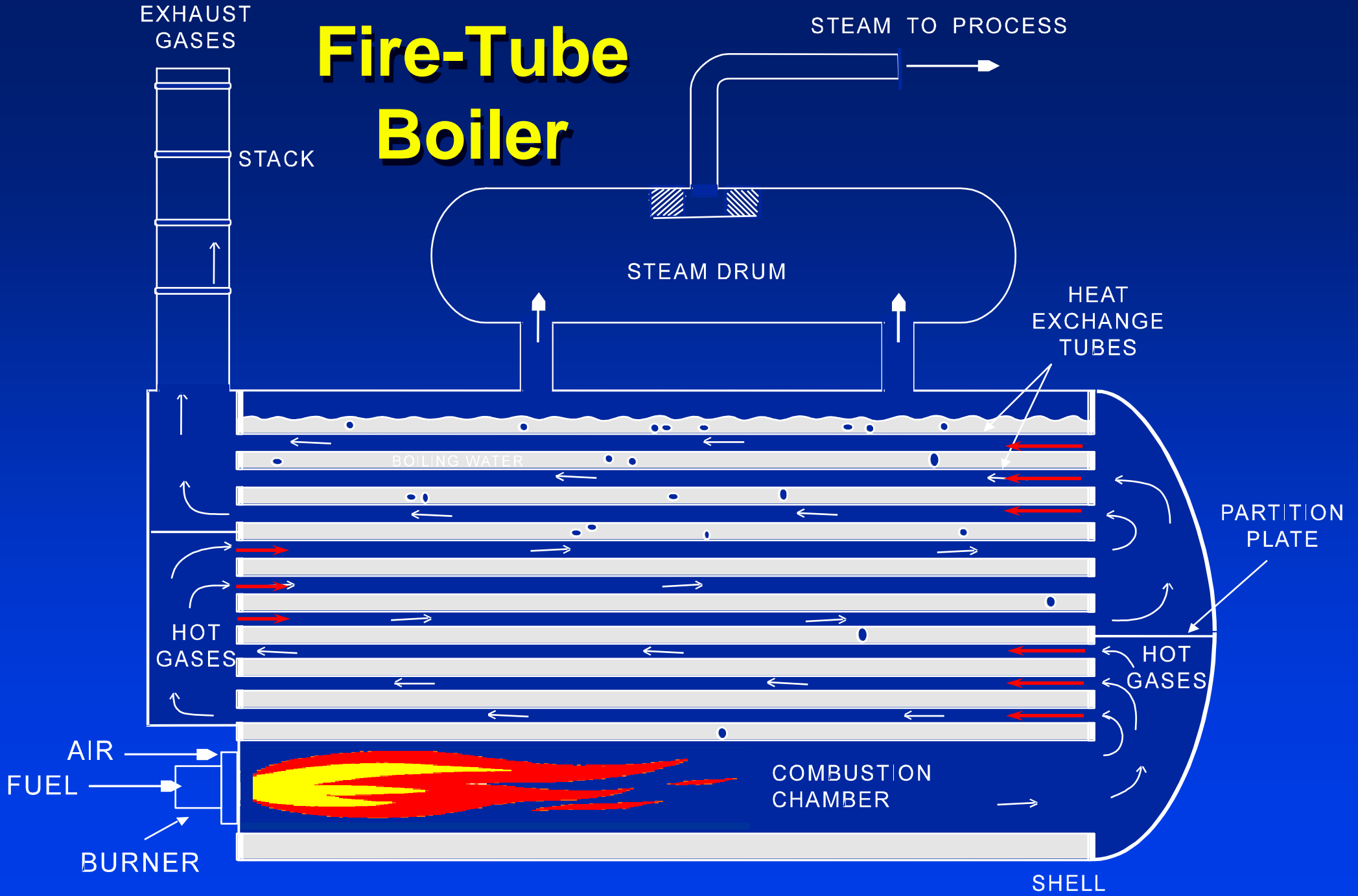
RATED STEAMING CAPACITY	1,540,000 LB/HR	MAX PRESSURE	2300 PSI	SUPERHEATER	955°F		
REHEATER CAPACITY	1,452,500 LB/HR	MAX PRESSURE	650 PSI	TEMPERATURE	855°F		
HEATING SURFACES	SQ. FT. BOILER	WATER WALLS	32,980	ECONOMIZER	115,000		
BUILT TO A S M E			RULES				
CONTRACT NO.	6468	MFR'S NO.	20942	TYPE	RRP	YEAR BUILT	1970

**Typical Boiler Rating**

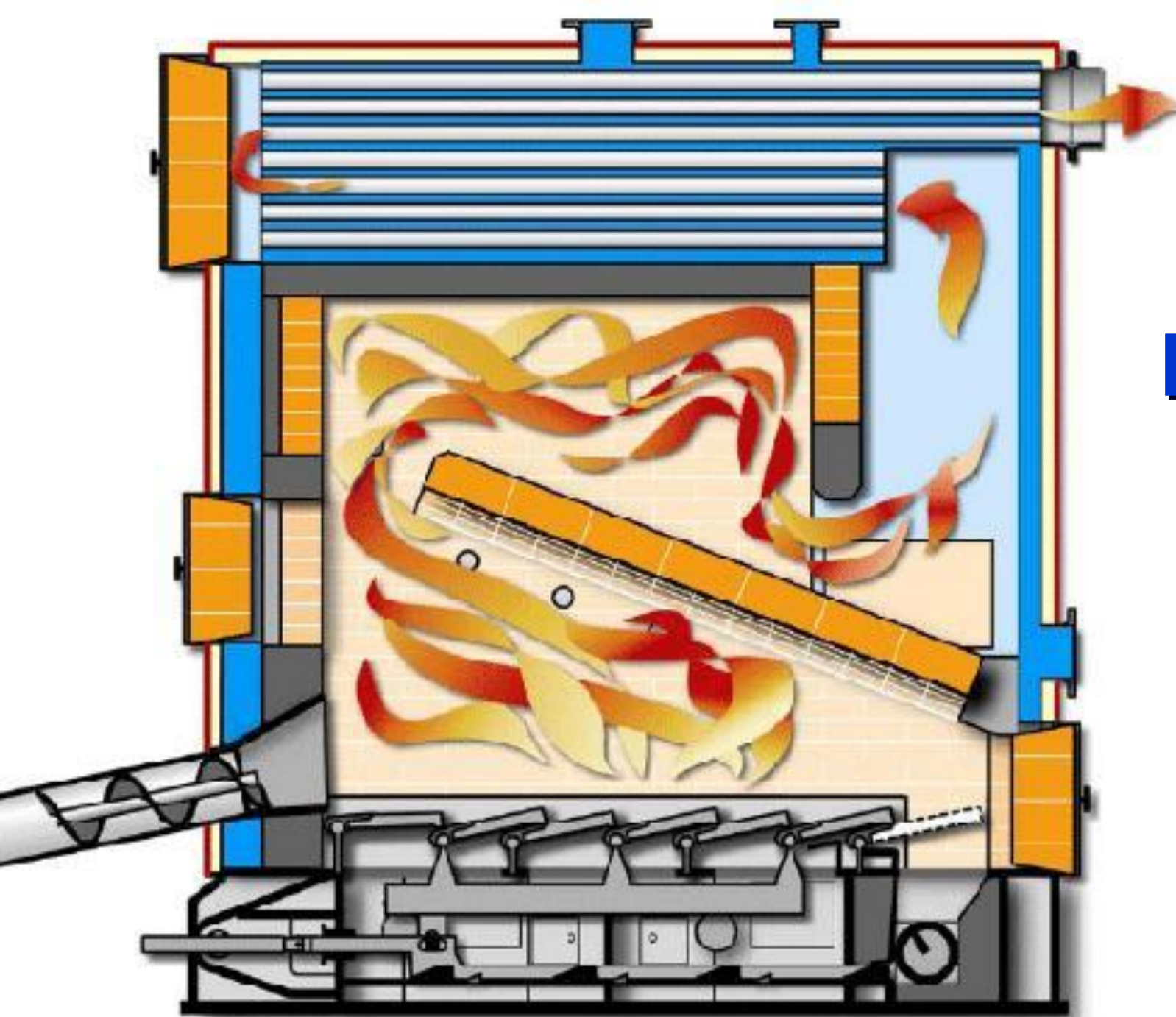


**Let's Discuss Firetube & Watertube Boilers**

# Fire-Tube Boiler

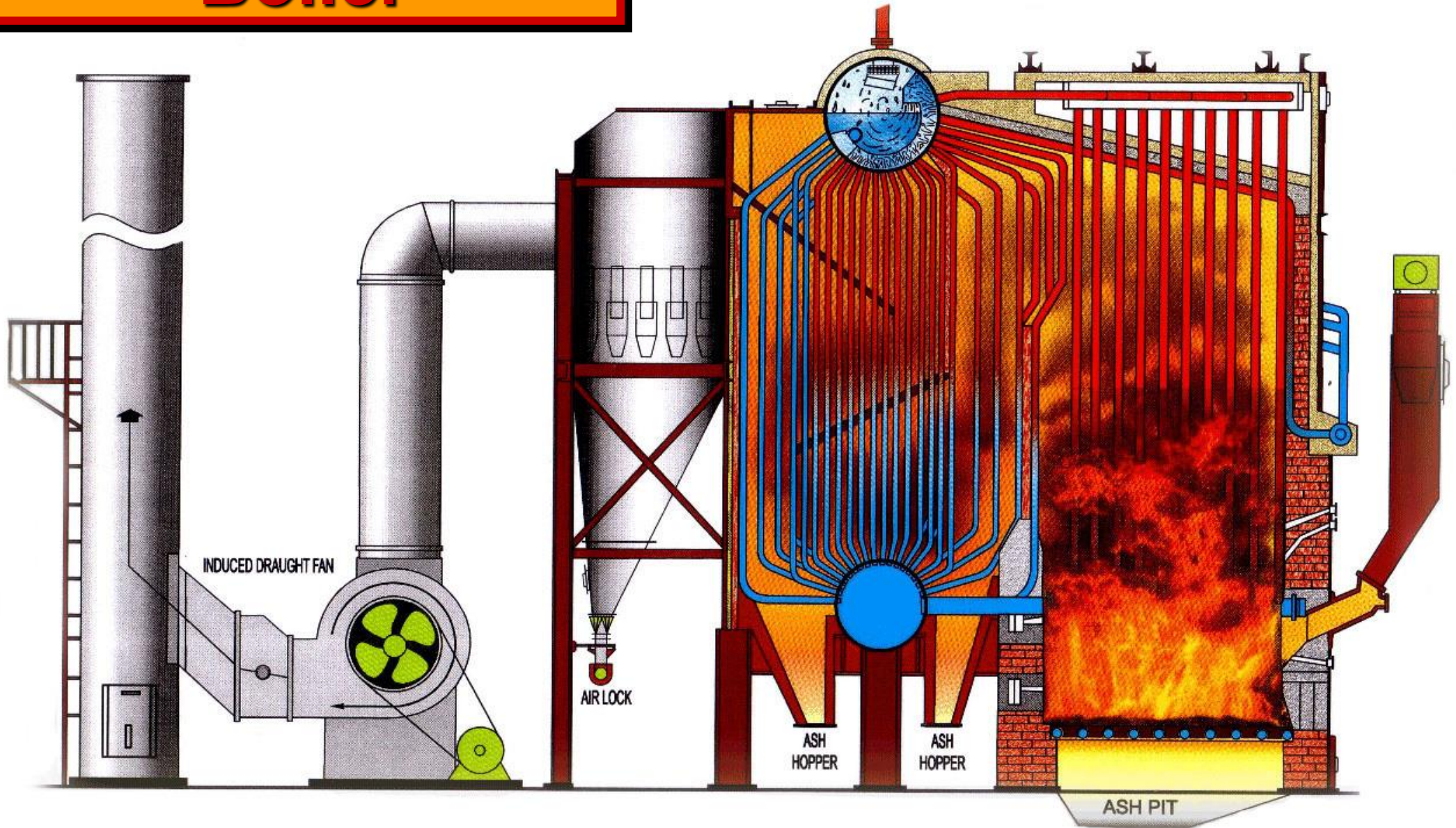






**Fire-Tube  
Boiler**

# Water-Tube Boiler





**Hybrid Boiler**



**Fire Wall**

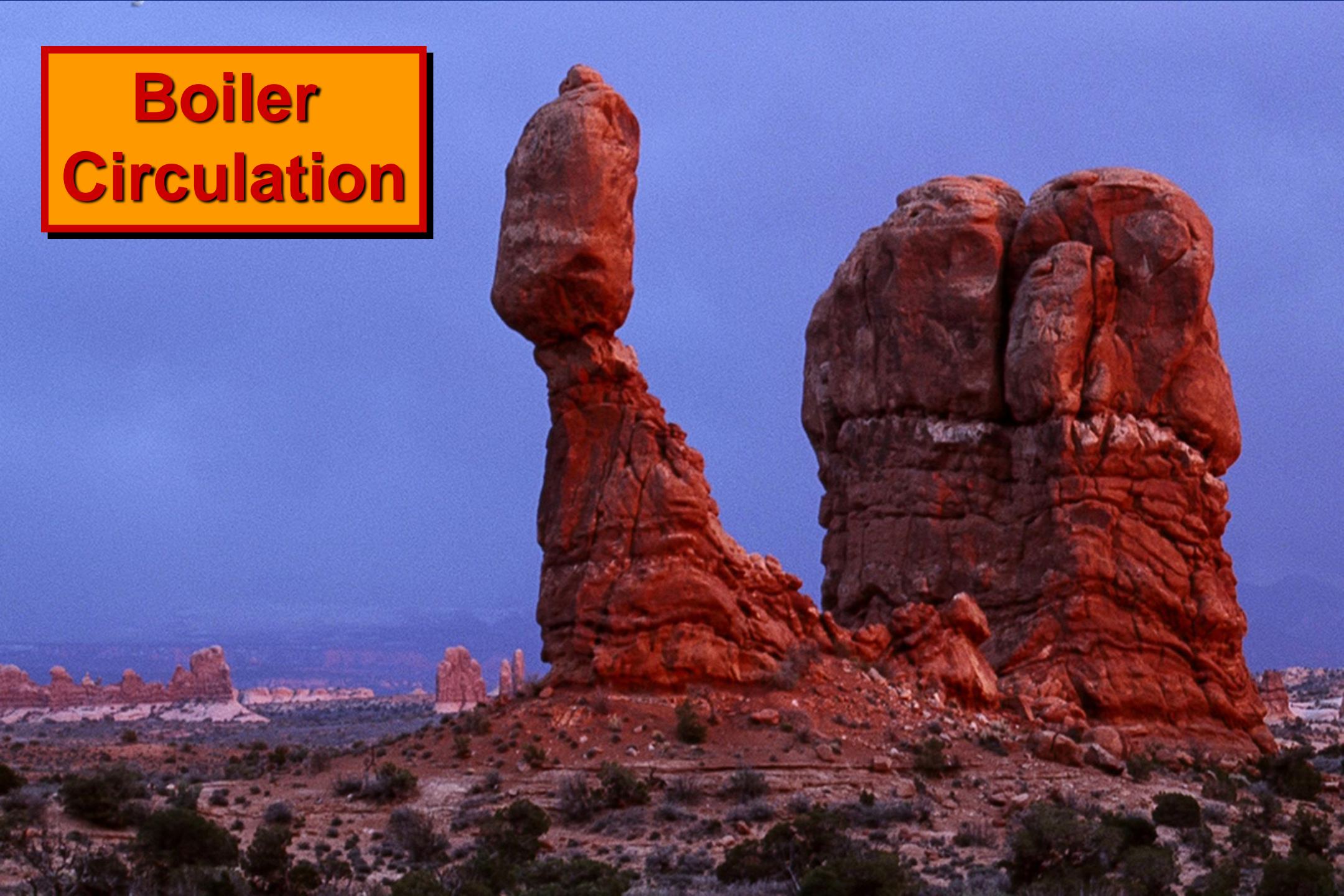


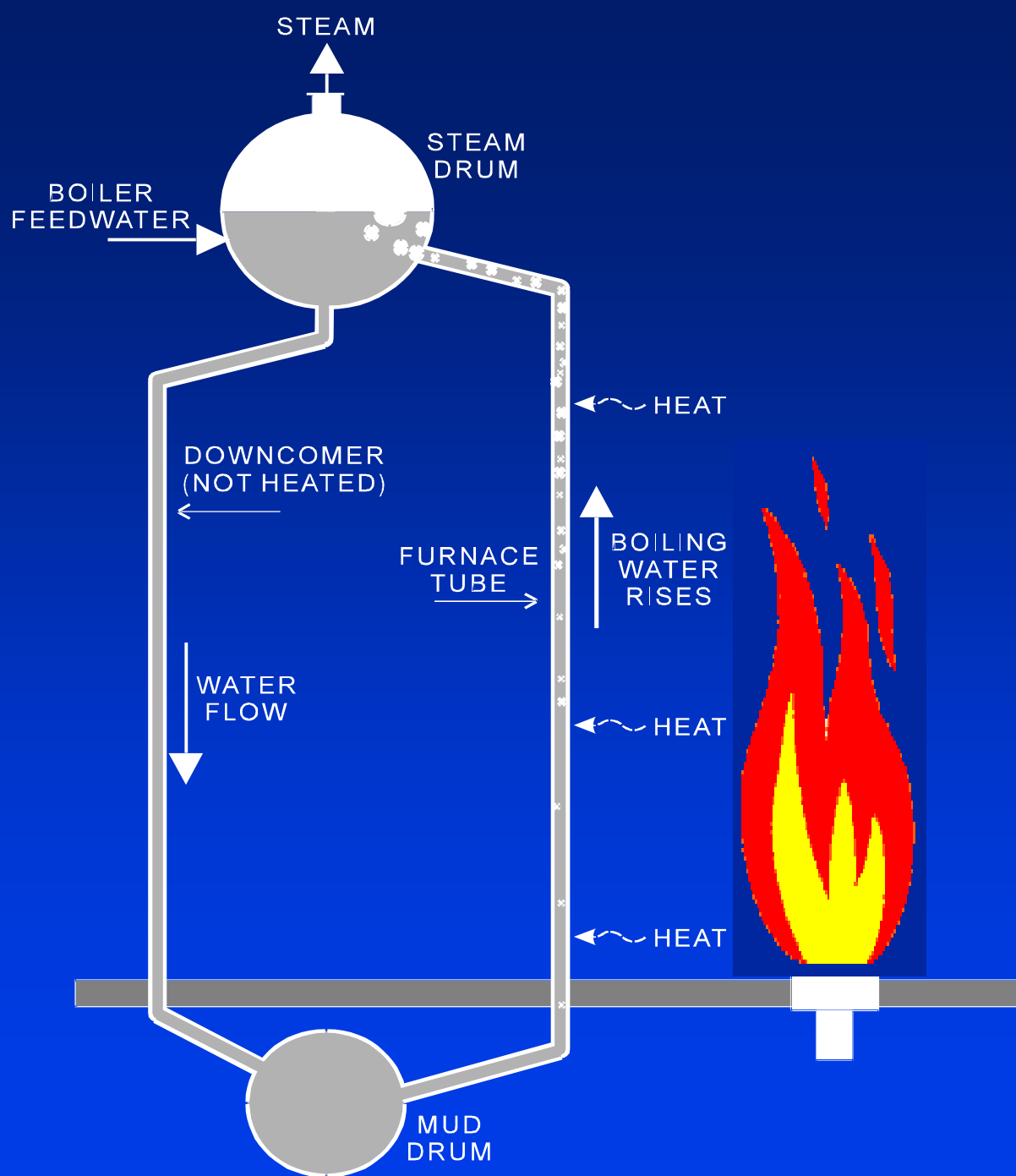
**Boiler Tubes**

The image shows a close-up, horizontal view of a stack of approximately ten boiler tubes. Each tube is covered in a dense, fine-grained mesh of small, rectangular fins, which are designed to increase the surface area for heat transfer. The tubes are stacked closely together, and the lighting highlights the texture of the mesh. A yellow rectangular box with a red border is overlaid at the bottom center of the image, containing the text "Boiler Tubes with Fins" in a bold, red, sans-serif font.

**Boiler Tubes with Fins**

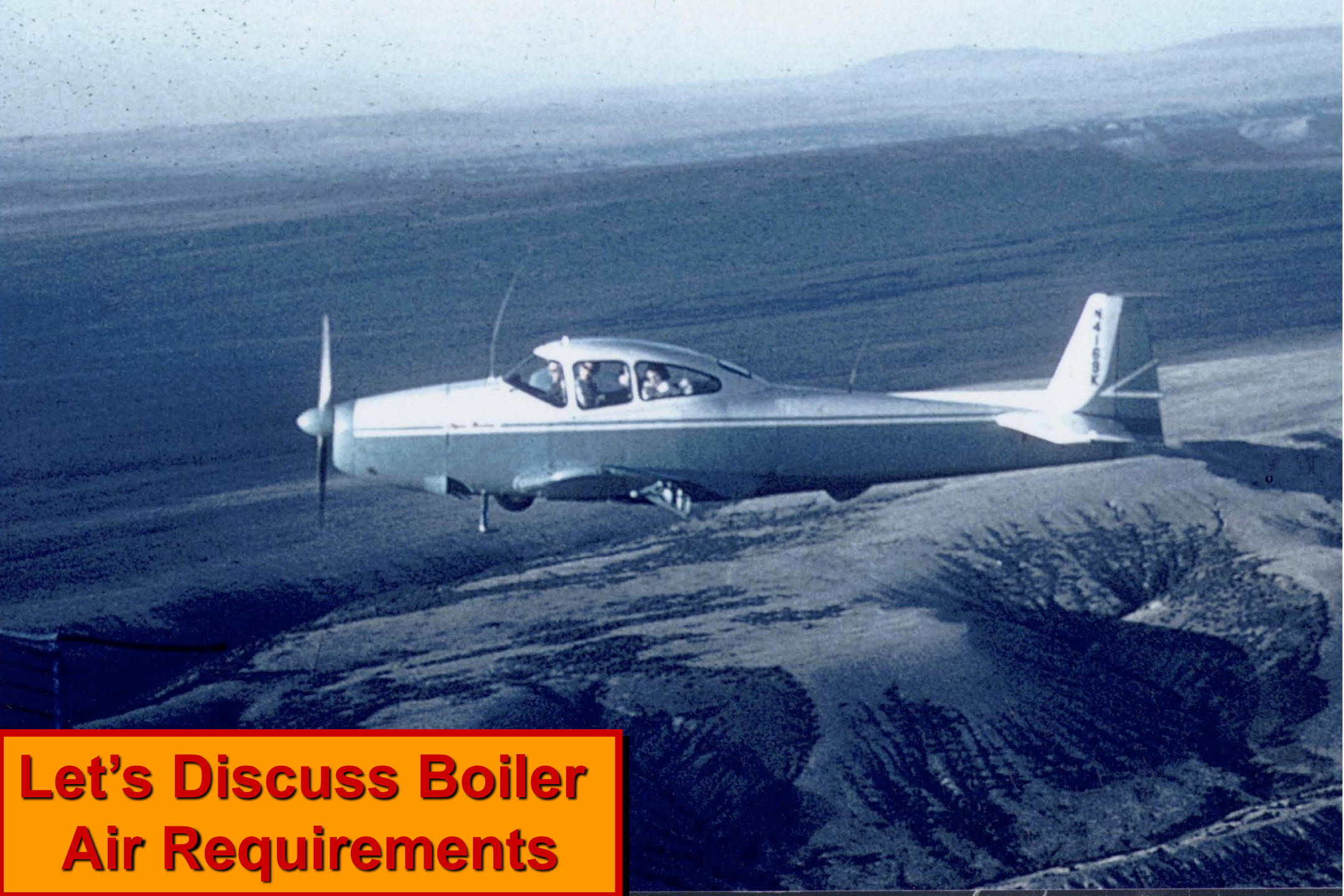
# Boiler Circulation





# Water to Steam Circulation Loop





**Let's Discuss Boiler  
Air Requirements**

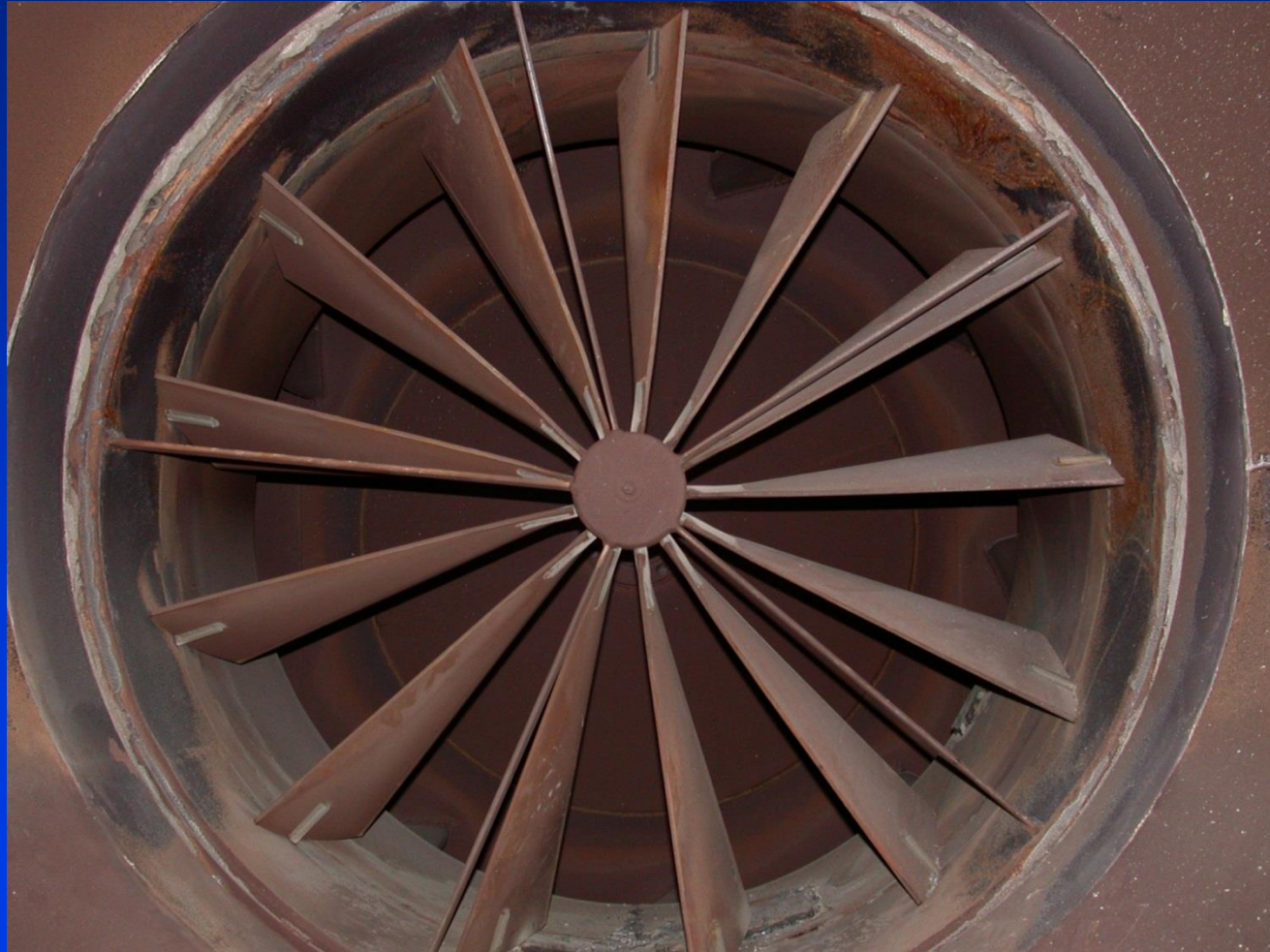
# Boiler Air Requirements

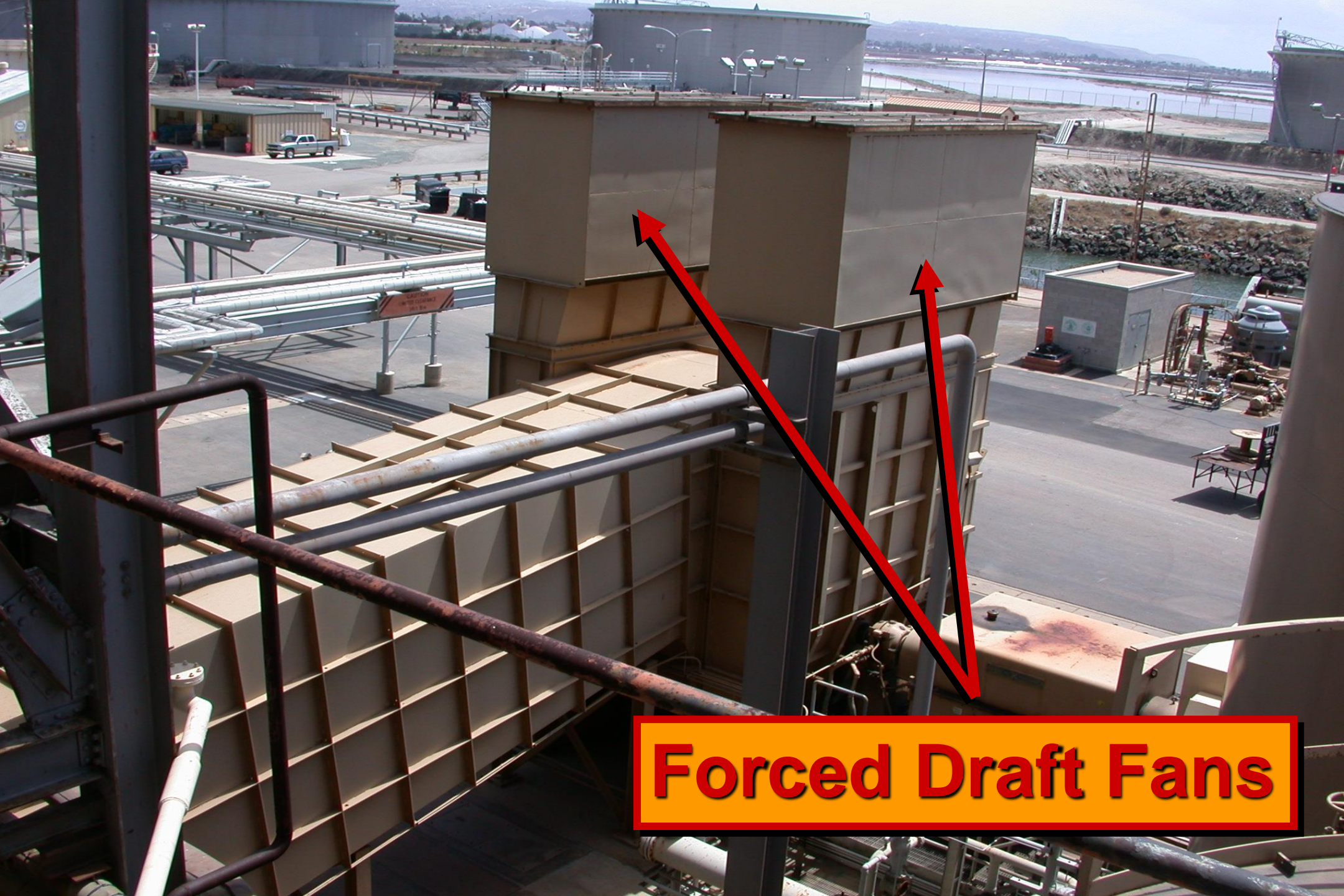
- ◆ **Draft**

- ◆ **Natural**
- ◆ **Forced**
- ◆ **Induced**

- ◆ **Combustion air**

- ◆ **Primary**
- ◆ **Secondary**
- ◆ **Excess**





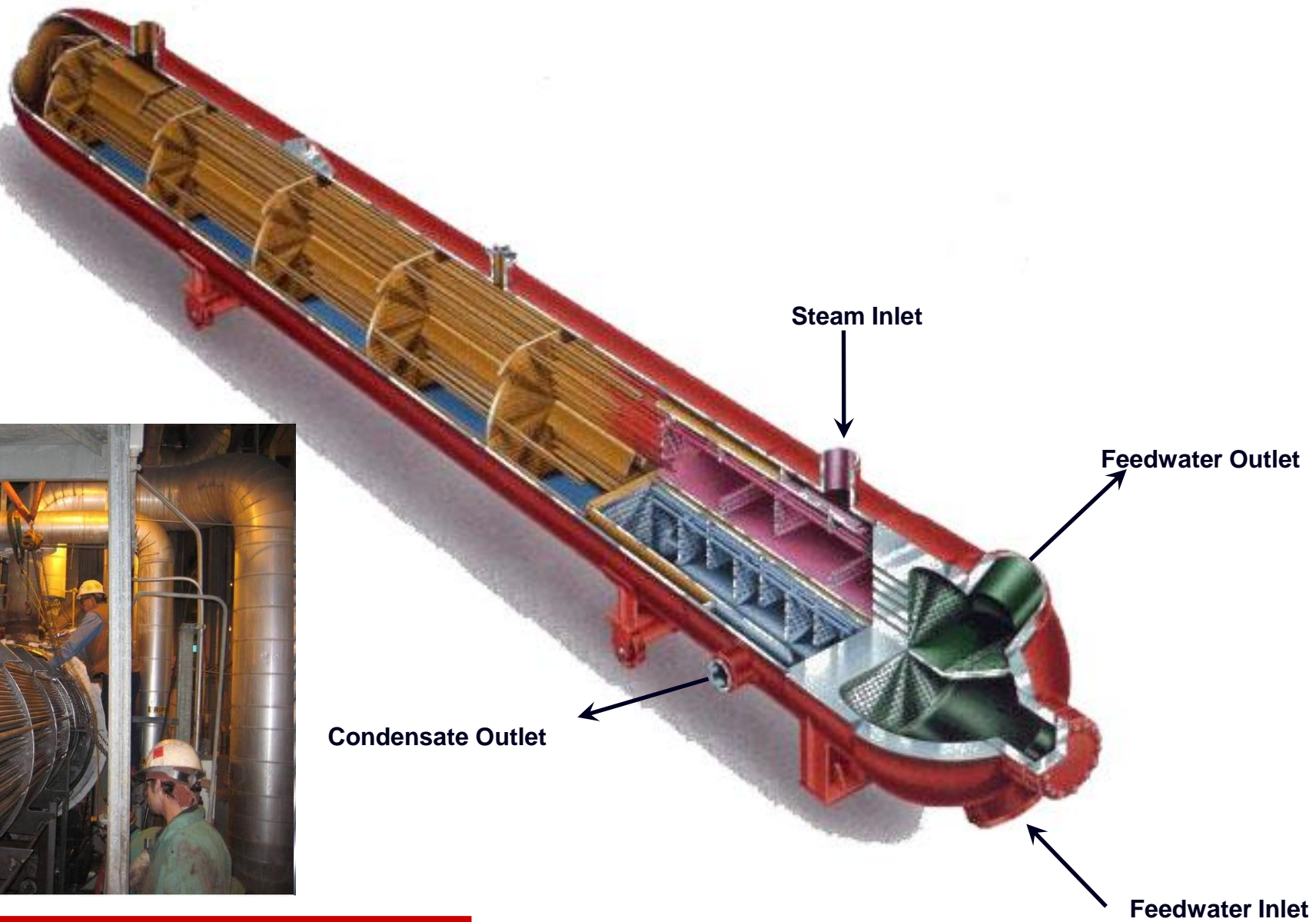
**Forced Draft Fans**

# Induced Draft Fans





**Let's Discuss Economizers,  
Feedwater Heaters & Air-Preheaters**

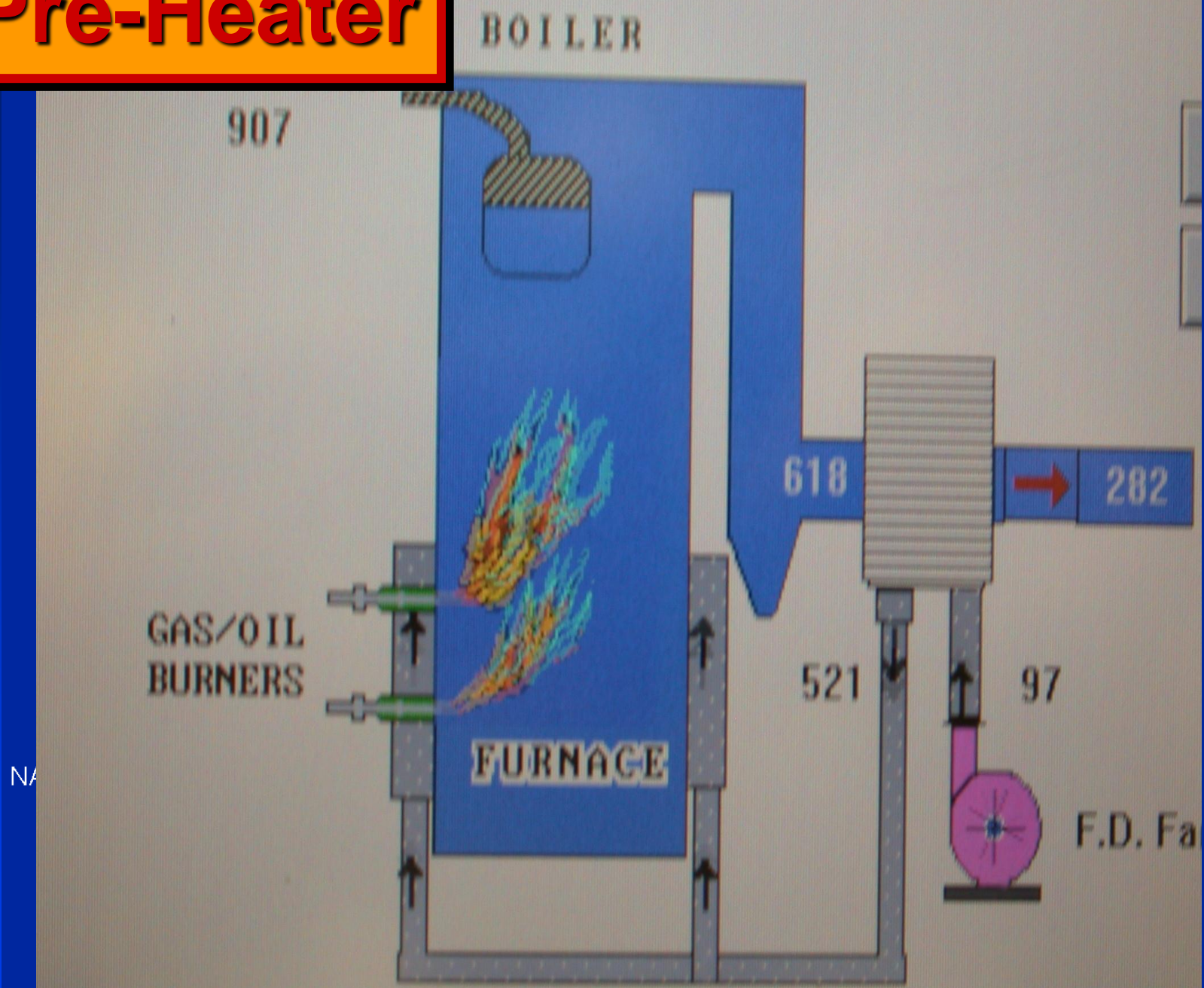


# Feedwater Heater

# Economizer – H<sub>2</sub>O Inside Tubes



# Air Pre-Heater

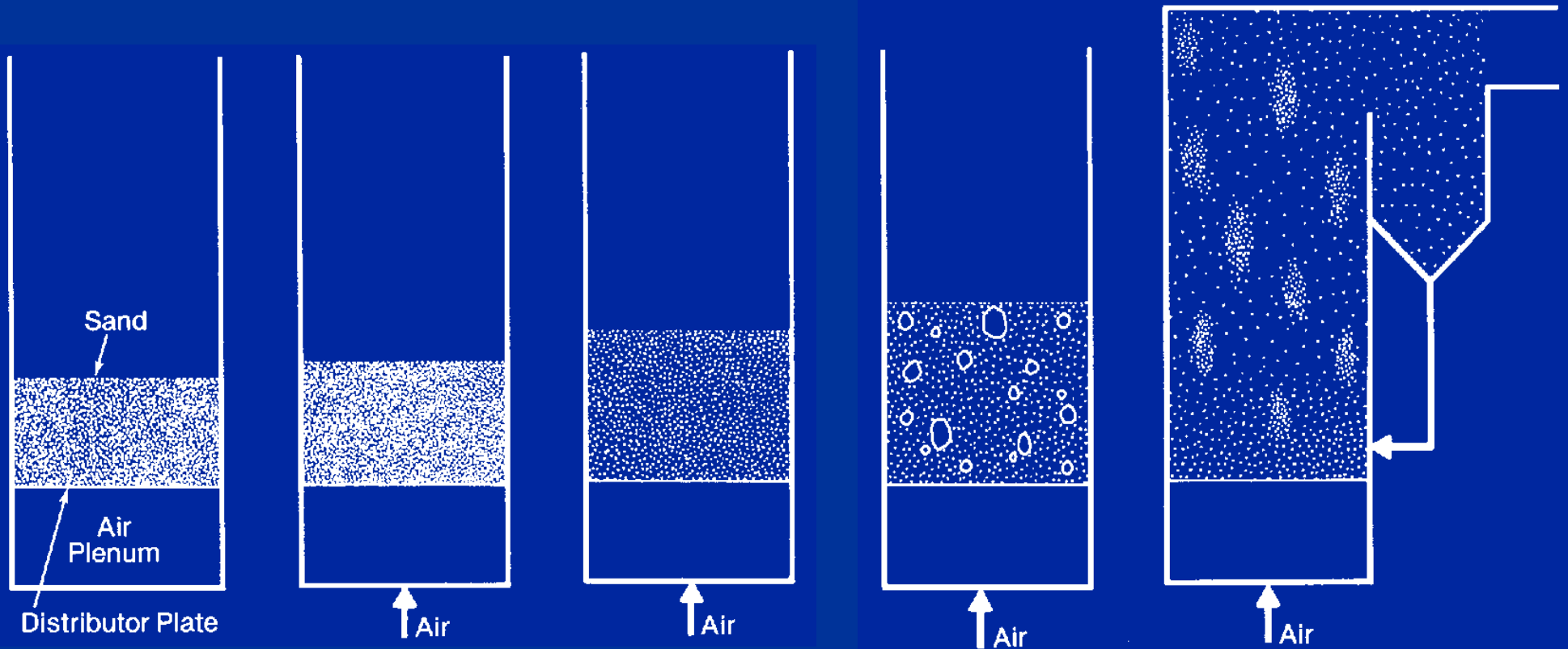






**Let's Discuss Fluidized  
Bed Boilers**

# Fluidized Bed Modes



Start  
No Air Flow

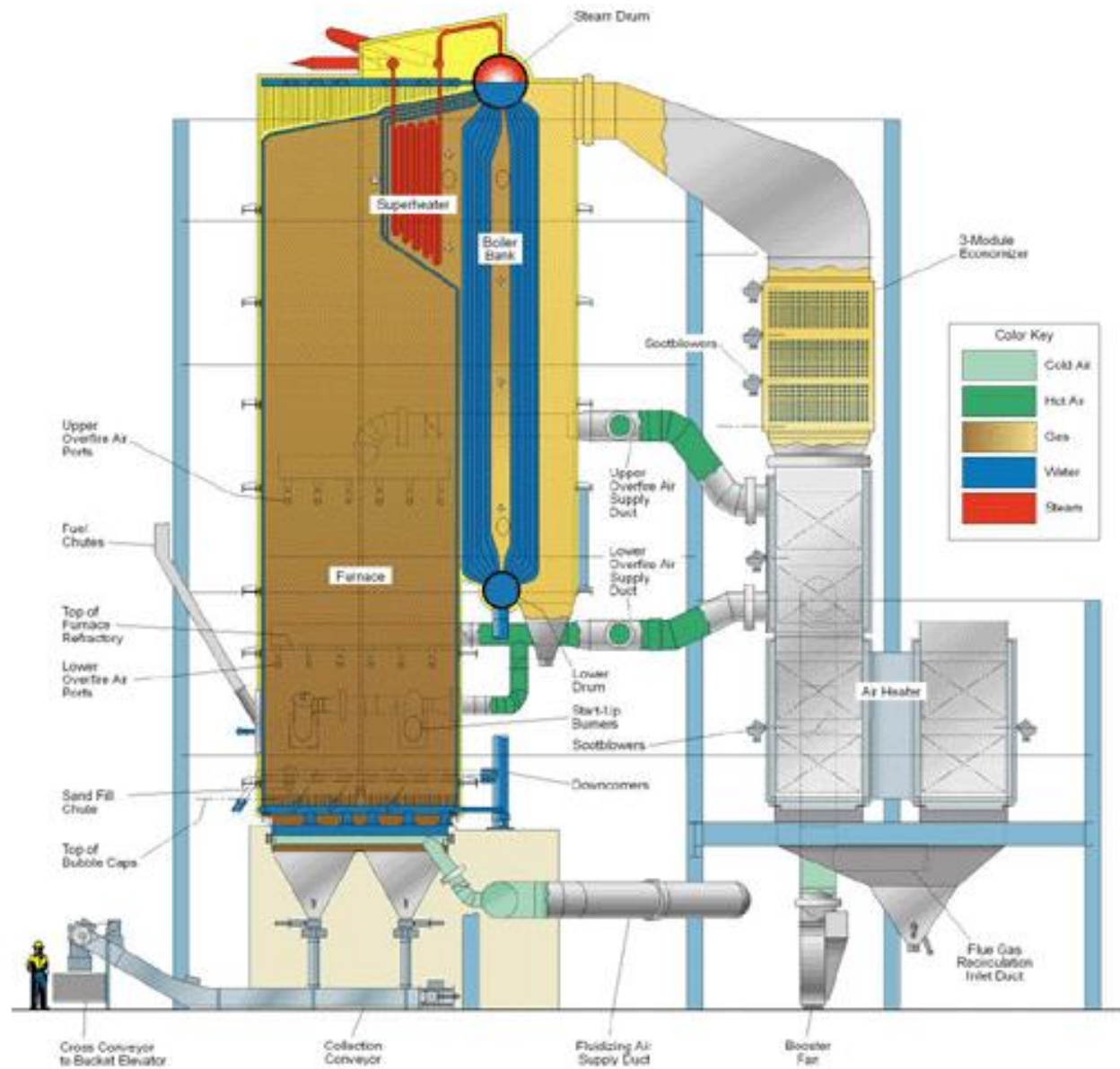
Fixed  
Bed

Minimum  
Fluidization

Bubbling  
Bed

Circulating  
Bed

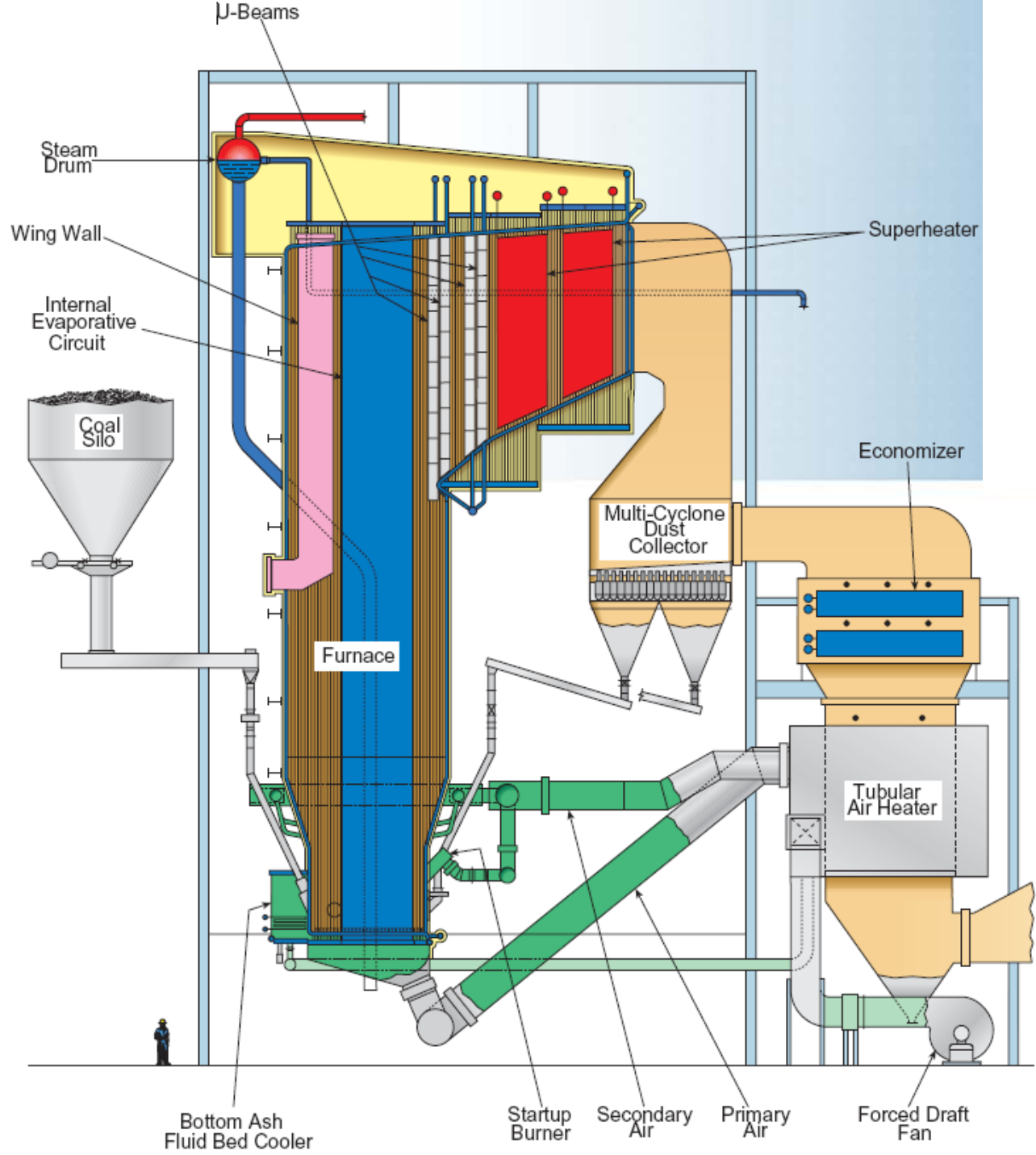
## BFB Bottom Supported



**Bubbling Fluidized Bed (BFB)**



**Circulating Fluidized Bed (CFB) Boilers**



# Circulating Fluidized Bed Boiler



**Superheater**

**Economizer**

**Furnace**

**Air Preheater**

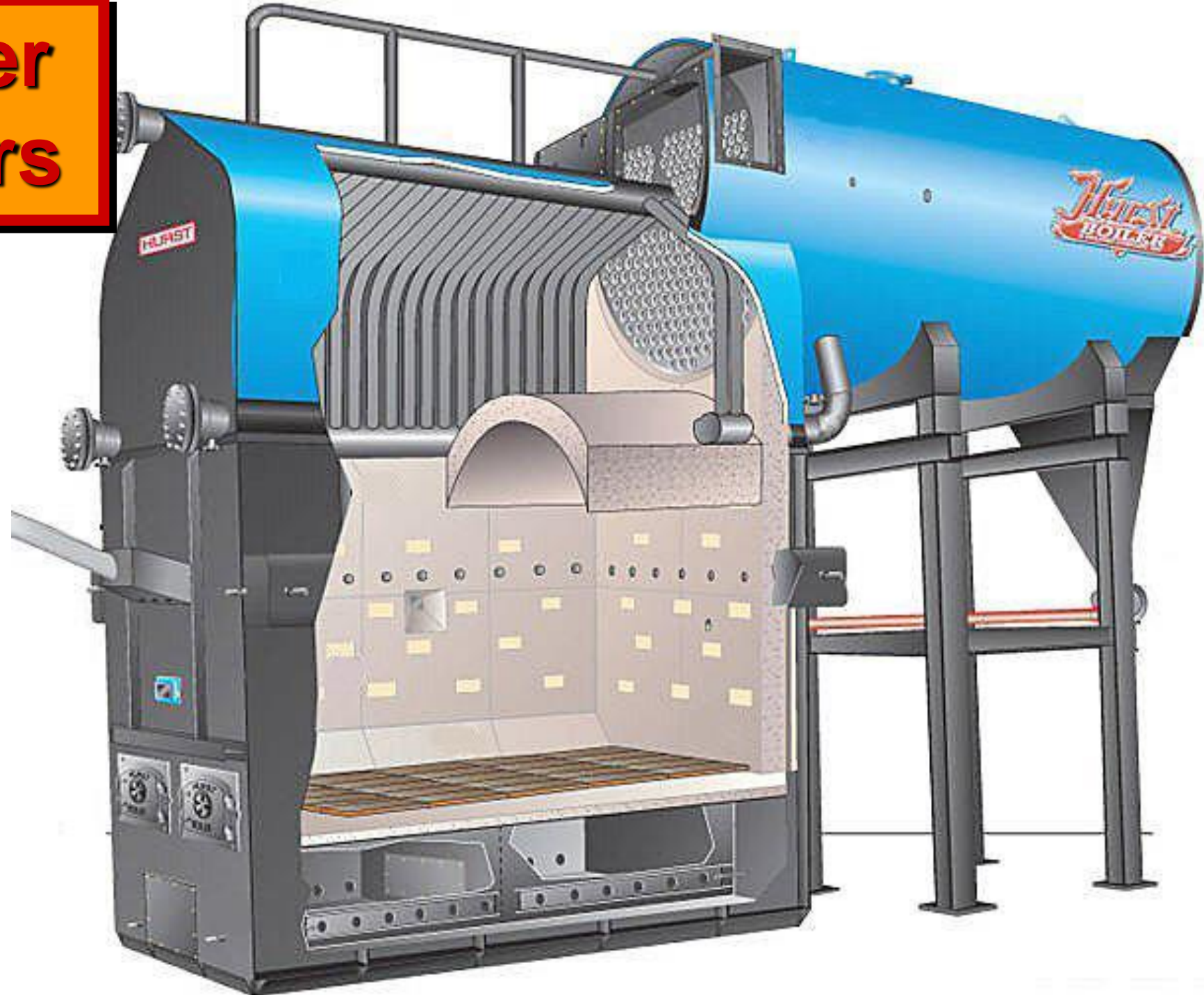
# Let's Discuss Stoker Boilers

I'm a **STOKER**  
jack and I'm Ok

I sleep all night  
and I work all day



# Stoker Boilers





# Stoker Boilers

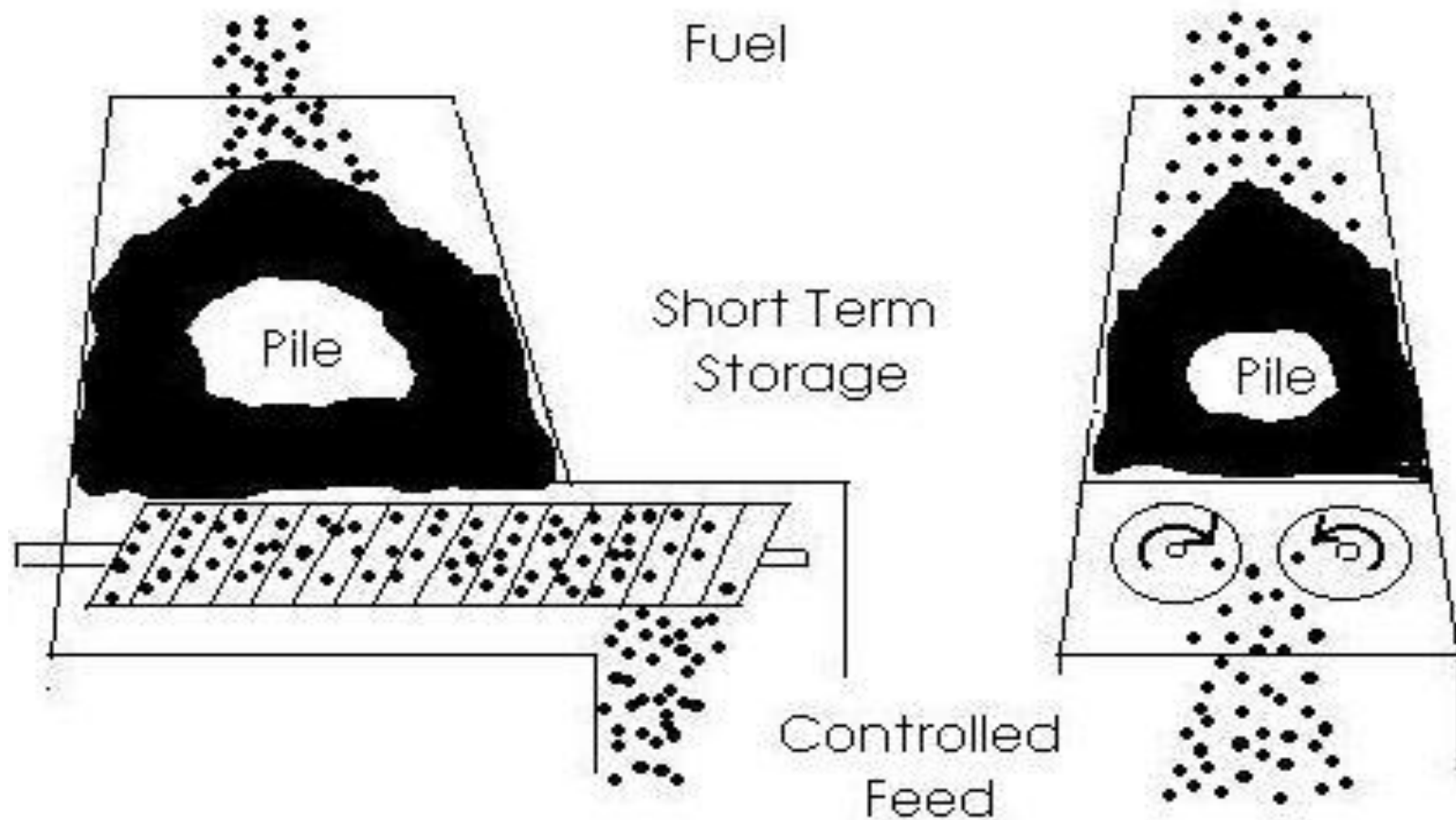


# ***Combustion Using a Stoker Boiler***

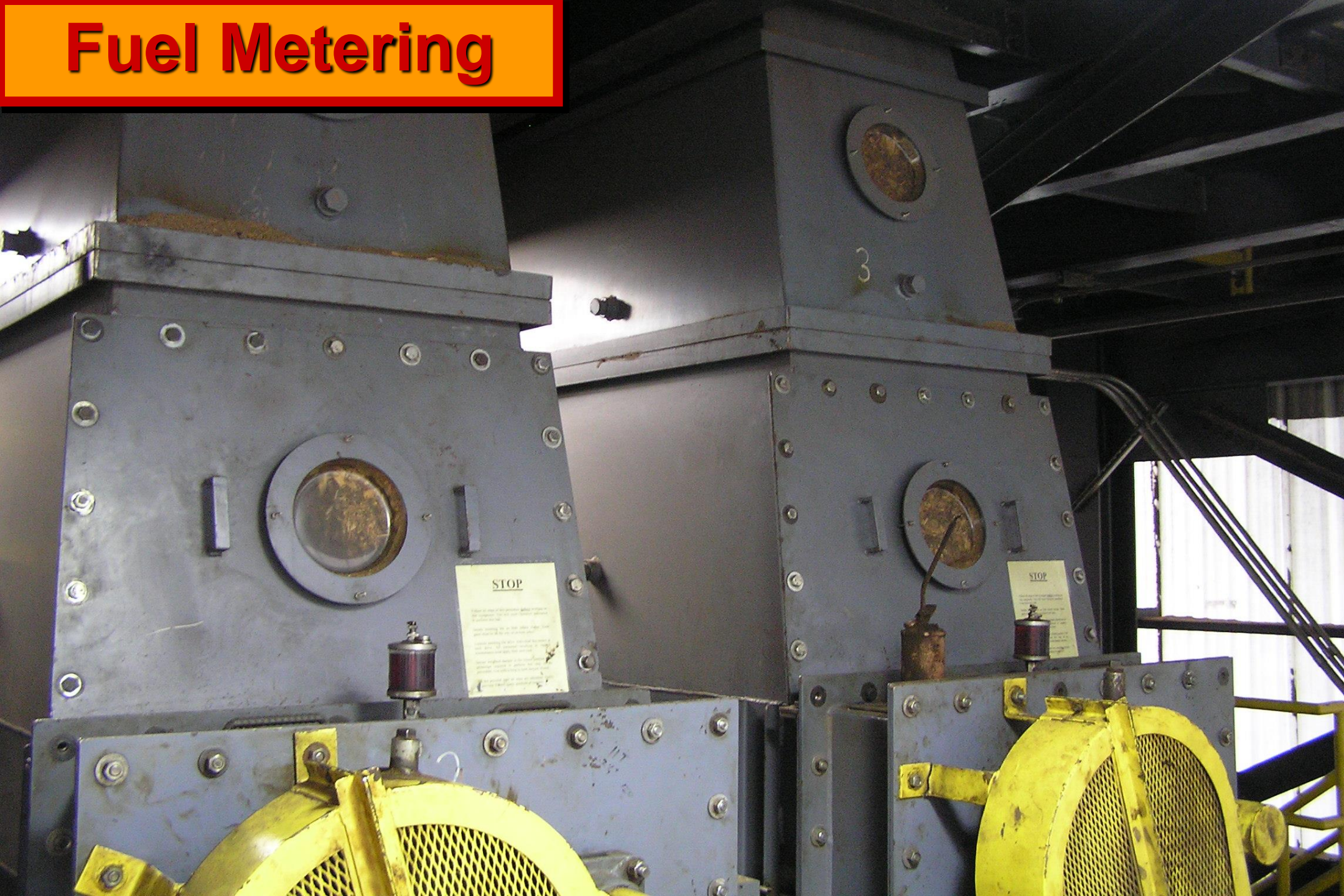
- ◆ **“Stoker” involves combustion on a grate**
- ◆ **Fuel Distribution Onto the Grate**
- ◆ **Undergrate or Underfire Air**
- ◆ **Overfire Air**
- ◆ **Three Steps of Biomass Combustion**
  - ◆ **Step 1 - Drying**
  - ◆ **Step 2 - Gasification and Volatile burnout**
  - ◆ **Step 3 - Char Burnout (Step #3) on the grate**



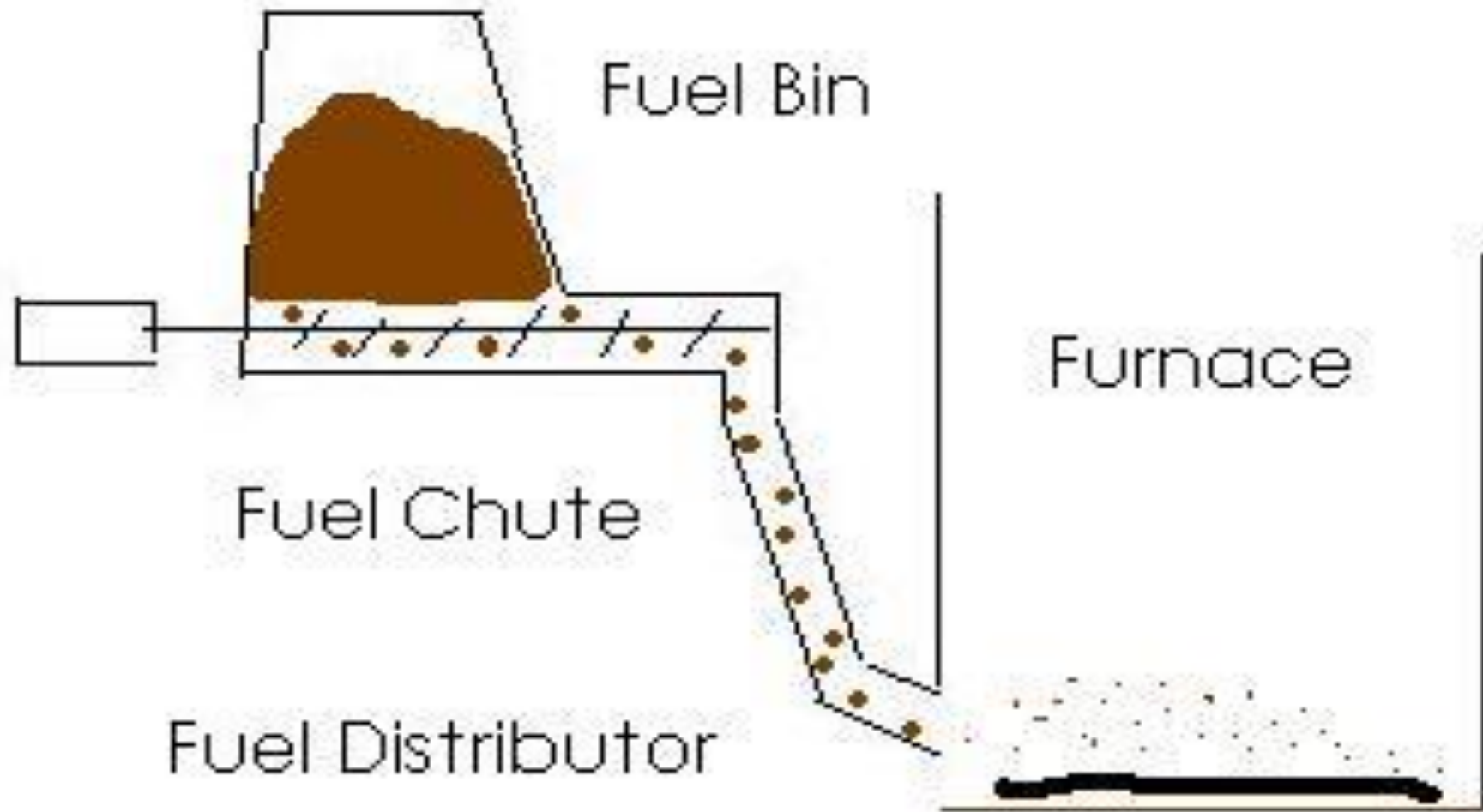
# Fuel Metering



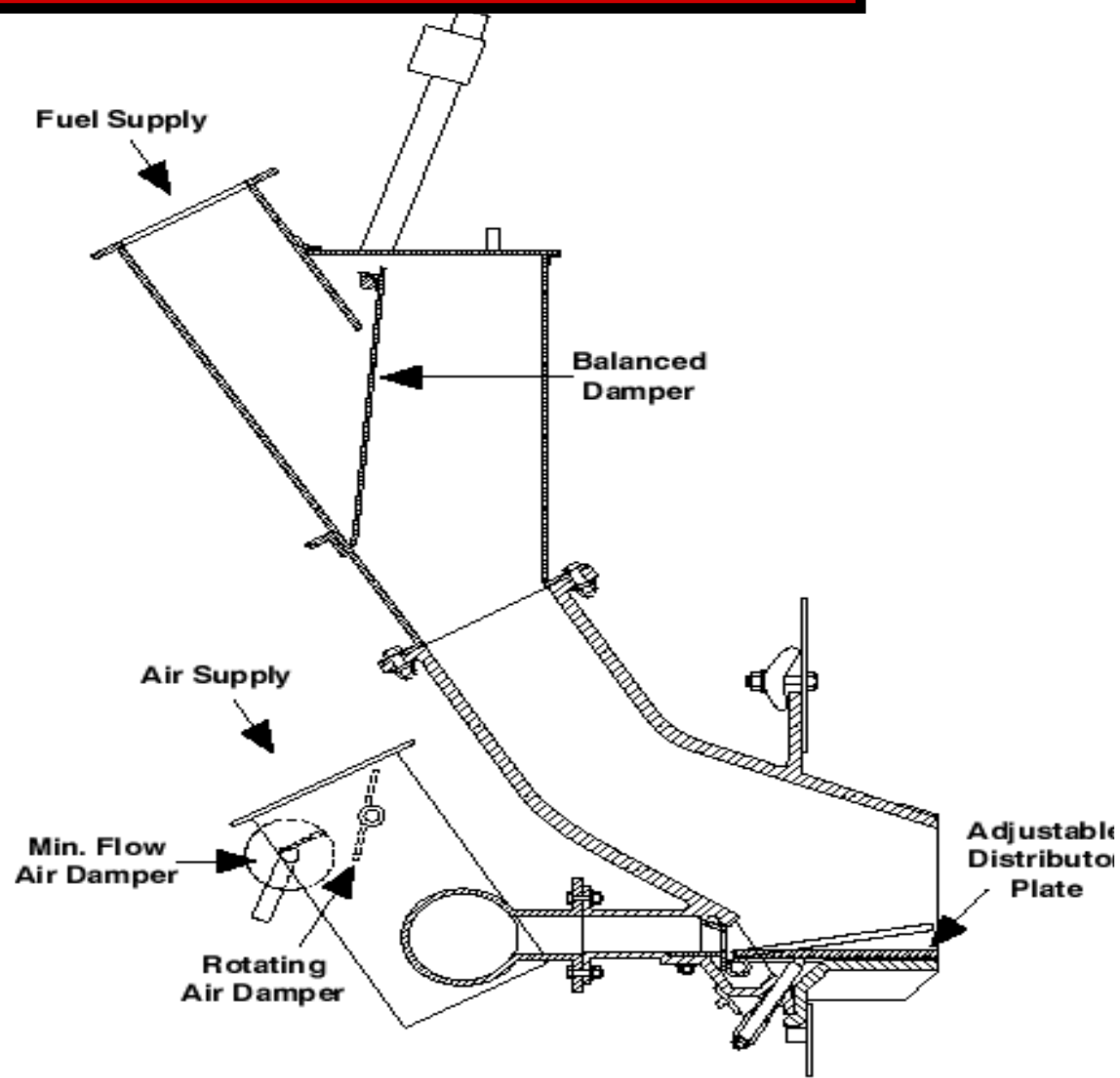
# Fuel Metering



# Fuel Chute and Simple Distribution



# Pneumatic Distribution



**Detroit Air Swept Fuel Distributor  
Model AD**

# Pneumatic Distribution

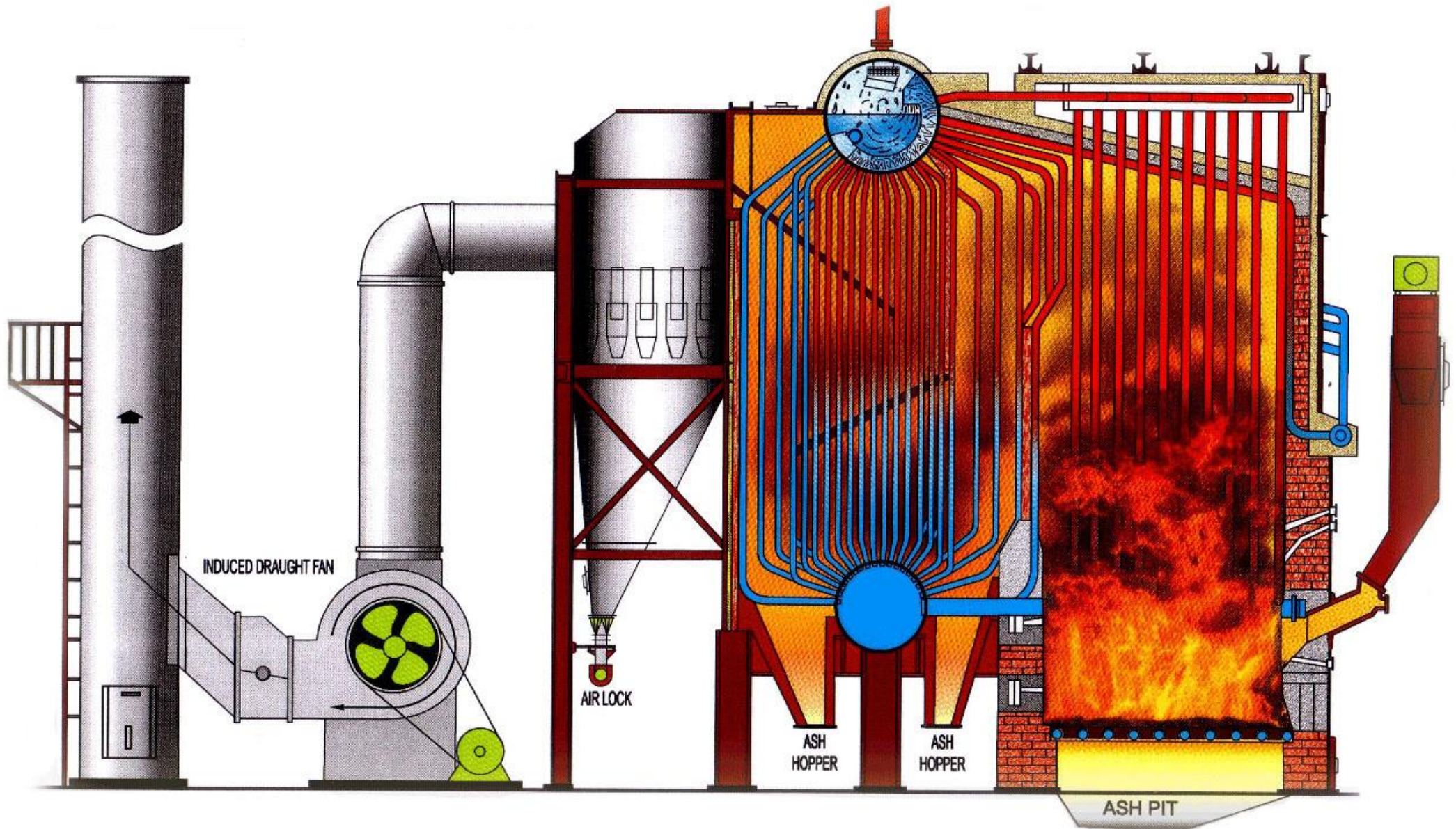


# Classified by Grate Designs

- ◆ Fixed Grate (Pinhole)
- ◆ Vibrating Grate
- ◆ Watercooled Hydrograte
- ◆ Reciprocating Grate
- ◆ Kablitz Grate
- ◆ Traveling Grate



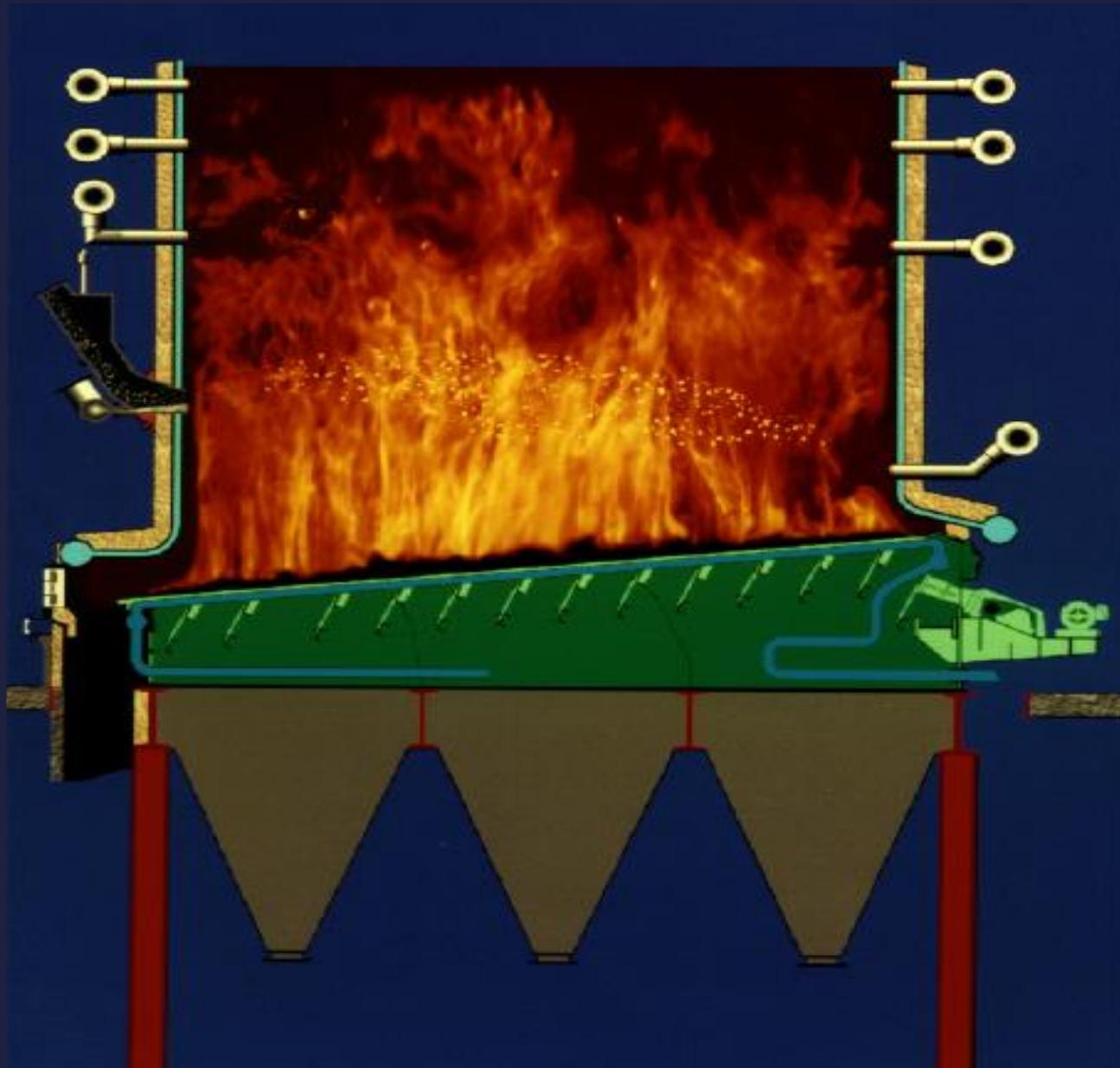
# Fixed Grate



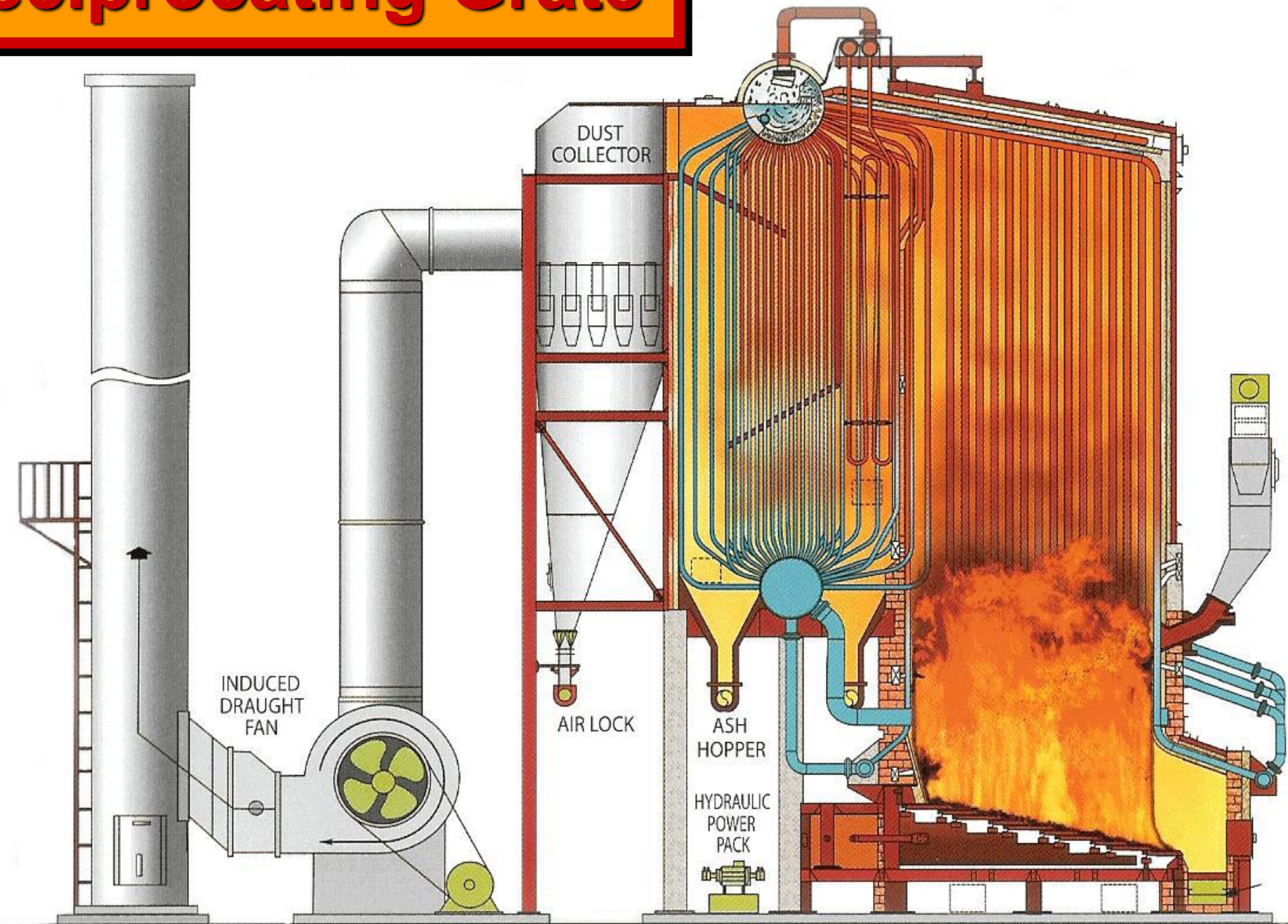
# Vibrating Grate



# *Watercooled Hydrograte*



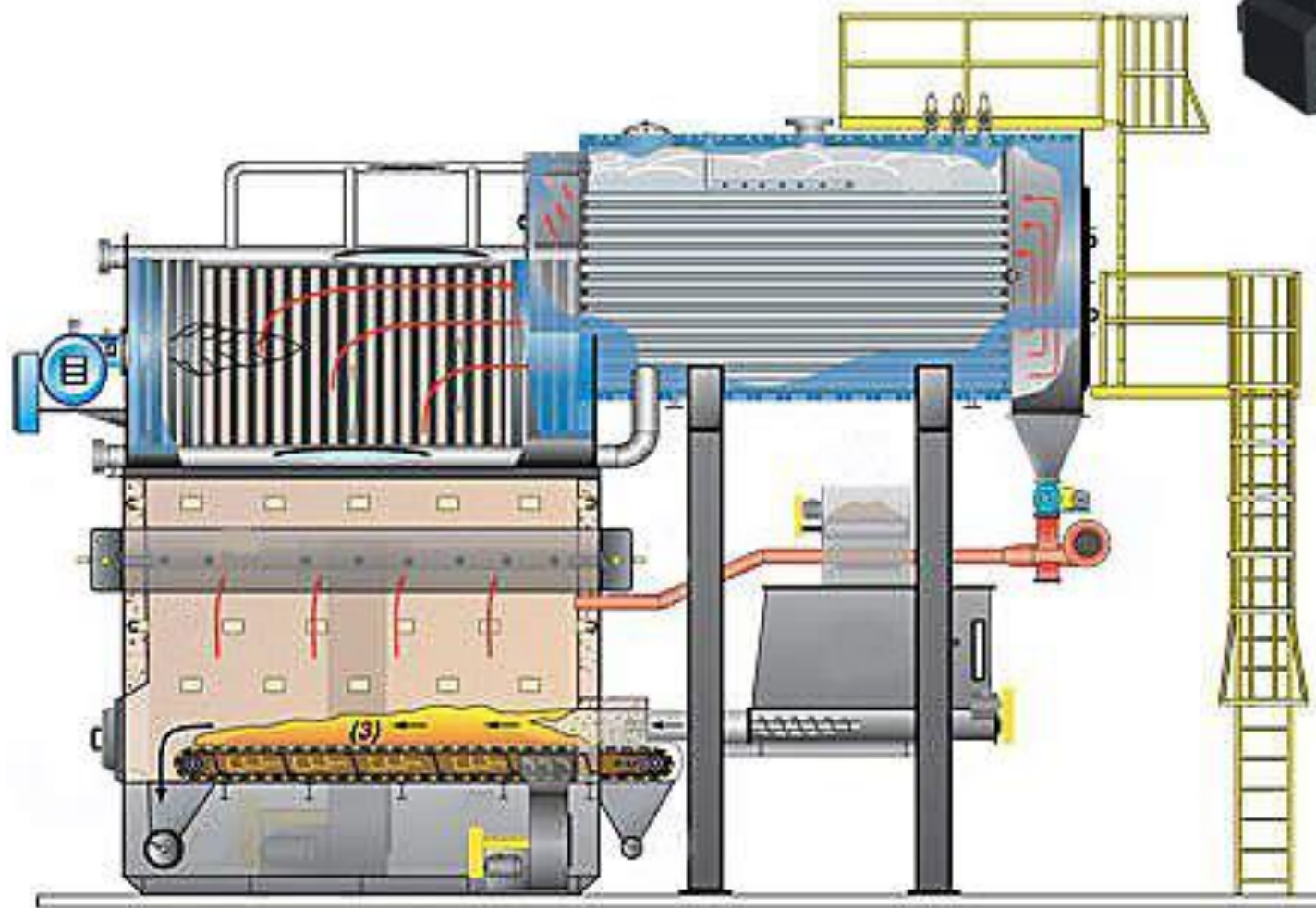
# Reciprocating Grate



# Kablitz Grate



# Traveling Grate



# Traveling Grate

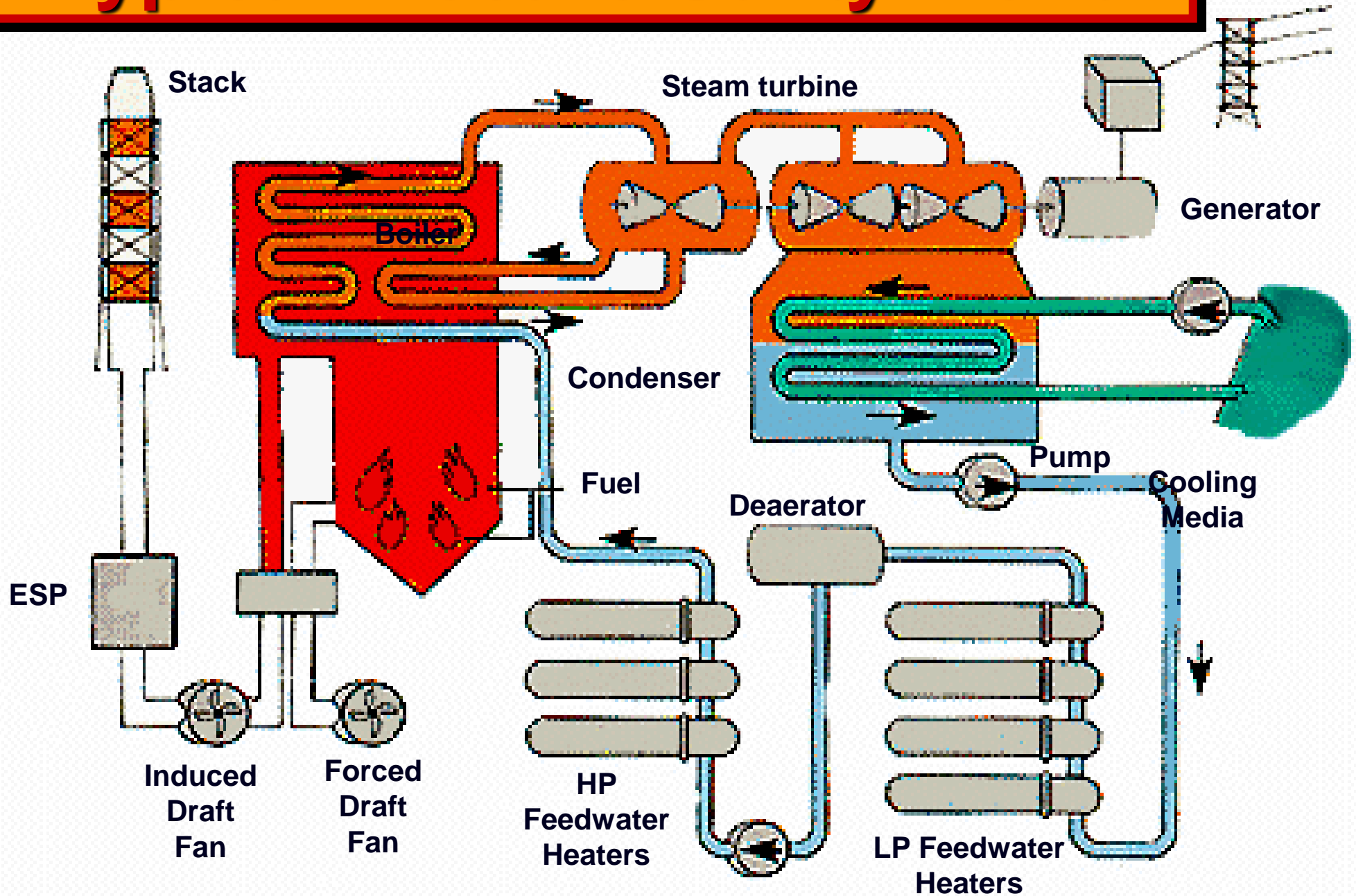


# Let's Discuss Power Generation



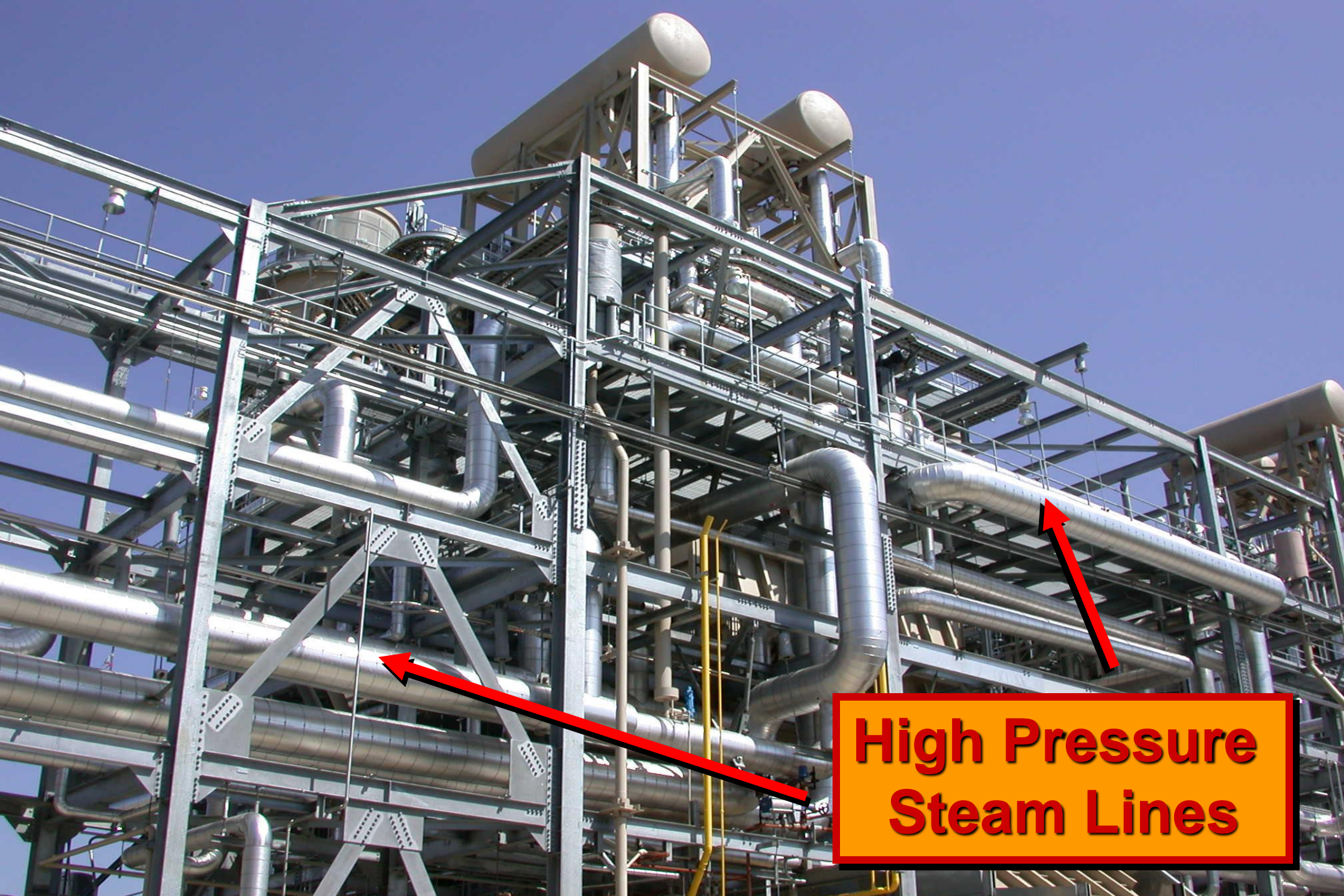


# Typical Electric Utility Plant



# High Pressure Steam Risers





**High Pressure  
Steam Lines**

# Steam Turbine



# Steam Turbine





**Generator**

**Steam Turbine**



# Cooling Towers

# Typical Control Room







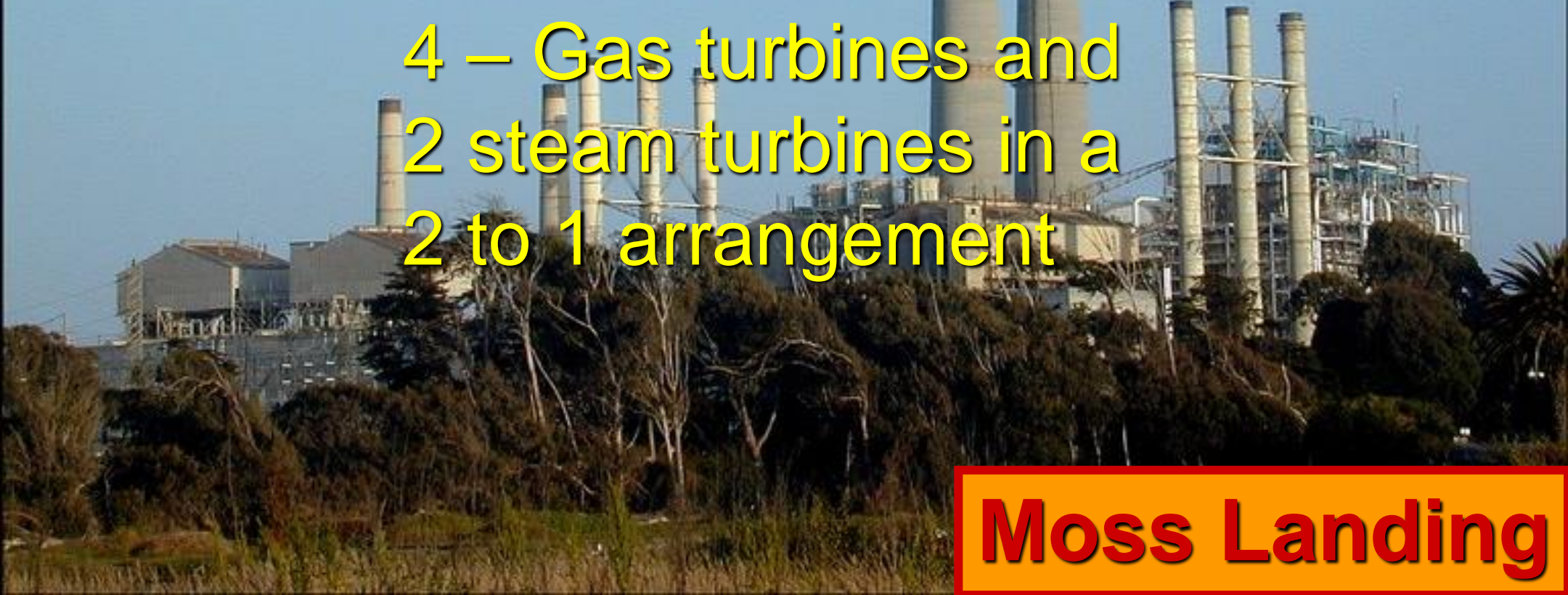
2,600 MW

2 – 750 MW boilers

4 – Gas turbines and

2 steam turbines in a

2 to 1 arrangement



**Moss Landing**

# Let's Discuss Biomass





**Waste Wood**



**Almond Shells**



**Urban Wood Waste**



**Walnut Shells**



**Orchard Trimmings**



**Corn Stover**



**Forest Debris**



**Rice Hulls**



**Biomass Fuels**


# Wood As A Fuel



- ◆ **Wood is man's oldest fuel**
- ◆ **Until very recently, wood was considered industrially as a waste material to be disposed of**
- ◆ **Escalating fuel costs and environmental concerns have changed things**
- ◆ **Wood use has opened up other "biofuels"**

# *History*

- ◆ 200,000 to 300,000 years - Controlled Use of Fire
- ◆ 10,000 to 20,000 years – Domesticated (living area)
- ◆ 800 years ago - First wood fuel shortages
- ◆ 400 years ago - Coal use in Europe
- ◆ 250 years ago - Industrial Revolution
- ◆ 150 years ago - Oil use
- ◆ 130 years ago - Natural gas use and electrification
- ◆ 70 years ago - Industrial wood-firing
- ◆ 50 years ago - Air pollutant investigations
- ◆ 20 years - Biofuels and “space age” investigations

- 
- ◆ **Wood differs from conventional fossil fuels**
    - ◆ **Physical structure**
    - ◆ **Chemical structure**
    - ◆ **Moisture content**
    - ◆ **“It’s alive” when harvested**
  - ◆ **Therefore it burns and must be burnt differently**
  - ◆ **Commercially viable for other uses**
  - ◆ **Relatively scarce resource**
  - ◆ **Development of “urban wood wastes”**

# Wood Fuel Physical Characteristics

- ◆ Xylem interior “white wood”
  - ◆ Board lumber
  - ◆ Chips
- ◆ Cambium layer
  - ◆ new growth
  - ◆ source of nutrients
- ◆ Bark



# Wood Fuel Physical Characteristics

- ◆ Debarking removes the Cambium layer and bark
- ◆ Yard wastes
- ◆ Processed through a hammermill (“hog”) for size reduction
- ◆ White wood
  - ◆ Sawdust
  - ◆ Sanderdust
  - ◆ Chip fines
- ◆ Other - plywood trim





# Waste Wood at Lumber Mill



**Mixed Fuel**

**Pallets, Urban Wood Waste, Tree Trimmings**



**Rice Hull Silo**



# Wood Fuel Variability

- ◆ **Random reclaim operations will result in significant variations in fuel quality**
- ◆ **Emphasizes the importance of good fuel management and blending**
- ◆ **Selective fuel purchasing is also very important**
- ◆ **But ... generally, you burn what you got**

# Fuel Quality

- ◆ **Poor combustion performance and high CO**
- ◆ **Small wood particles**
  - ◆ **Not handled well in the feed and distribution systems**
  - ◆ **Rapidly entrained with insufficient time to complete the 3-Step combustion process within the furnace**
- ◆ **High moisture content**
  - ◆ **Tends to pile on the grate**
  - ◆ **Causes “thick-bed” grate conditions**
  - ◆ **Disrupts undergrate airflow**

# Fuel Preparation

- ◆ Screening
- ◆ Metal Removal
- ◆ Drying
- ◆ Deicing
- ◆ Sizing
- ◆ Blending



# Wood Fuel Sizing

- ◆ **“Overs” (> 3 inches) => Plug fuel chutes**
  - ◆ Screen out
  - ◆ Mill to a smaller size
- ◆ **“Fines” (<1/4 in) => Not completely burned**
  - ◆ Segregate
  - ◆ Blend up to 20% with larger materials

# Wood Fuel Screening





**Tub Grinder for  
Gross Size  
Reduction**



# Hammermill for Fine Size Reduction



# Wood Fuel Drying

- ◆ **Part of another wood processing operation**
  - ◆ **Kiln dried trim**
  - ◆ **Dried planer shavings**
  - ◆ **Sanderdust**
- ◆ **MC > 65% requires some type of drying or blending with drier fuel**
- ◆ **MC < 15% is potentially explosive**

# Fuel Blending



# Importance of Blending

- ◆ **Control moisture content**
- ◆ **Improve fines burnout**
- ◆ **Implement by:**
  - ◆ **Gross mixing using a front-end loader**
  - ◆ **Separate fuel bins feeding a common feed system**



# Wood Fuel Characteristics - Moisture Content

- ◆ Water needed for life
- ◆ Present in the cell structure and on surface
- ◆ Moisture content varies with
  - ◆ Species
  - ◆ Location
  - ◆ Season
  - ◆ Handling practices
- ◆ Nature levels of 30% to 65%+
- ◆ Kiln dried to less than 10%

# Effect of Moisture

- ◆ **Decreases combustion temperatures**
- ◆ **Leads to incomplete combustion and the generation of higher levels of CO and ash C**
- ◆ **Decreases boiler efficiency**
- ◆ **Leads to more fuel use, higher energy costs and increased air pollutants**

# Wood Fuel Characteristics - Volatility

- ◆ 70% to 80% of dry wood is “volatile” hydrocarbons
- ◆ Released from the wood structure at relatively low temperatures (500 F)
- ◆ Volatiles burn “in suspension” away from the wood particles
- ◆ Balance is “fixed carbon” or “char”

# Wood Chemistry

## Hydrocarbons

- ◆ Methane  $\text{CH}_4$
- ◆ Ethane  $\text{C}_2\text{H}_6$
- ◆ Propane  $\text{C}_3\text{H}_8$
- ◆ Complex Fuels  $\text{C}_x\text{H}_y$
- ◆ More Complex Fuels  $\text{C}_x\text{H}_y\text{S}_z\text{N}_a$



# Wood Species

Parameter	Pine	Redwood	Hemlock	Fir
C (%wt,dry)	50.3	53.5	50.4	52.3
H (%wt,dry)	6.2	5.9	5.8	6.3
O (%wt,dry)	43.1	40.3	41.4	40.5
N (%wt,dry)	0.04	0.10	0.10	0.10
S (%wt,dry)	< 0.1	< 0.1	< 0.1	< 0.1
Ash (%wt,dry)	0.3	0.2	2.2	0.8
Moisture (%)	30 - 60	30 - 60	30 - 60	30 - 60
Btu/lb	9,153	9,220	8,620	9,058

# **3-Step Combustion Process**

## **Step #1 - Drying**

- ◆ **Fresh wood particles absorb heat by convection and radiation from the ongoing combustion processes**
- ◆ **Initially, the heat energy evaporates the water in the cells and on the surface**
- ◆ **The evaporated water vapor diffuses away and mixes with the combustion products**

# **3-Step Combustion Process**

## **Step #2 – Devolatilization**

- ◆ **The wood fuel particles continue to absorb heat**
- ◆ **The energy absorbed releases the volatile combustibles**
- ◆ **The volatiles diffuse away, mix with air (oxygen), and burn**
- ◆ **The energy released radiates back and helps to sustain combustion**

# **3-Step Combustion Process**

## **Step #3 - Char Burnout**

- ◆ **The char remains after the volatiles have been released**
- ◆ **Char is primarily pure carbon and requires:**
  - ◆ **An extended time period to burn**
  - ◆ **Air (oxygen) transported to it**
- ◆ **Inert ash remains after the char burnout**



# Ash

- ◆ Intrinsic ash (potassium) can combine with alumina and silica to form a relatively low melting point temperature ash (1600 to 1800 F)
- ◆ Potassium also volatilizes at low temperatures (1500 F) and reacts with ash in the boiler flue gas stream contributing to fouling problems
- ◆ Ash accumulation causes
  - ◆ Airflow problems
  - ◆ High draft levels and fan horsepower
  - ◆ Reduced superheat temperatures
  - ◆ Boiler shutdowns



**Ash  
Loadout**

# Ash Loadout



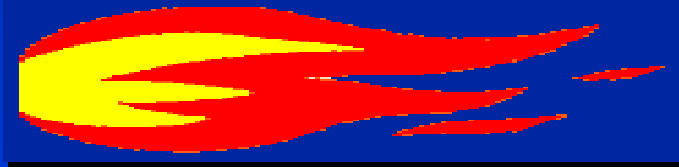




**Let's Discuss  
Emissions  
& Controls**

# Emissions From Boilers

**Fuel  
+  
Air  
(N<sub>2</sub>, O<sub>2</sub>)**



- ◆ H<sub>2</sub>O
- ◆ CO<sub>2</sub>
- ◆ CO
- ◆ NO<sub>x</sub>
- ◆ HC
- ◆ SO<sub>x</sub>
- ◆ PM
- ◆ Cl-

# Emissions Control Methods

- ◆ **Boiler design**
- ◆ **Proper maintenance**
- ◆ **Operating conditions**
- ◆ **Fuel types**
- ◆ **Combustion modifications**
- ◆ **Exhaust treatment**



# Control of Gaseous Emissions

- ◆ Low-NOx burners
- ◆ OFA
- ◆ Ammonia injection (SNCR)
- ◆ Catalysts (SCR)
- ◆ RSCR
- ◆ FGR



# Combustion Considerations

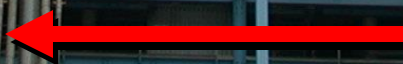
- ◆ Time
- ◆ Temperature
- ◆ Turbulence
- ◆ Oxygen
- ◆ Nitrogen



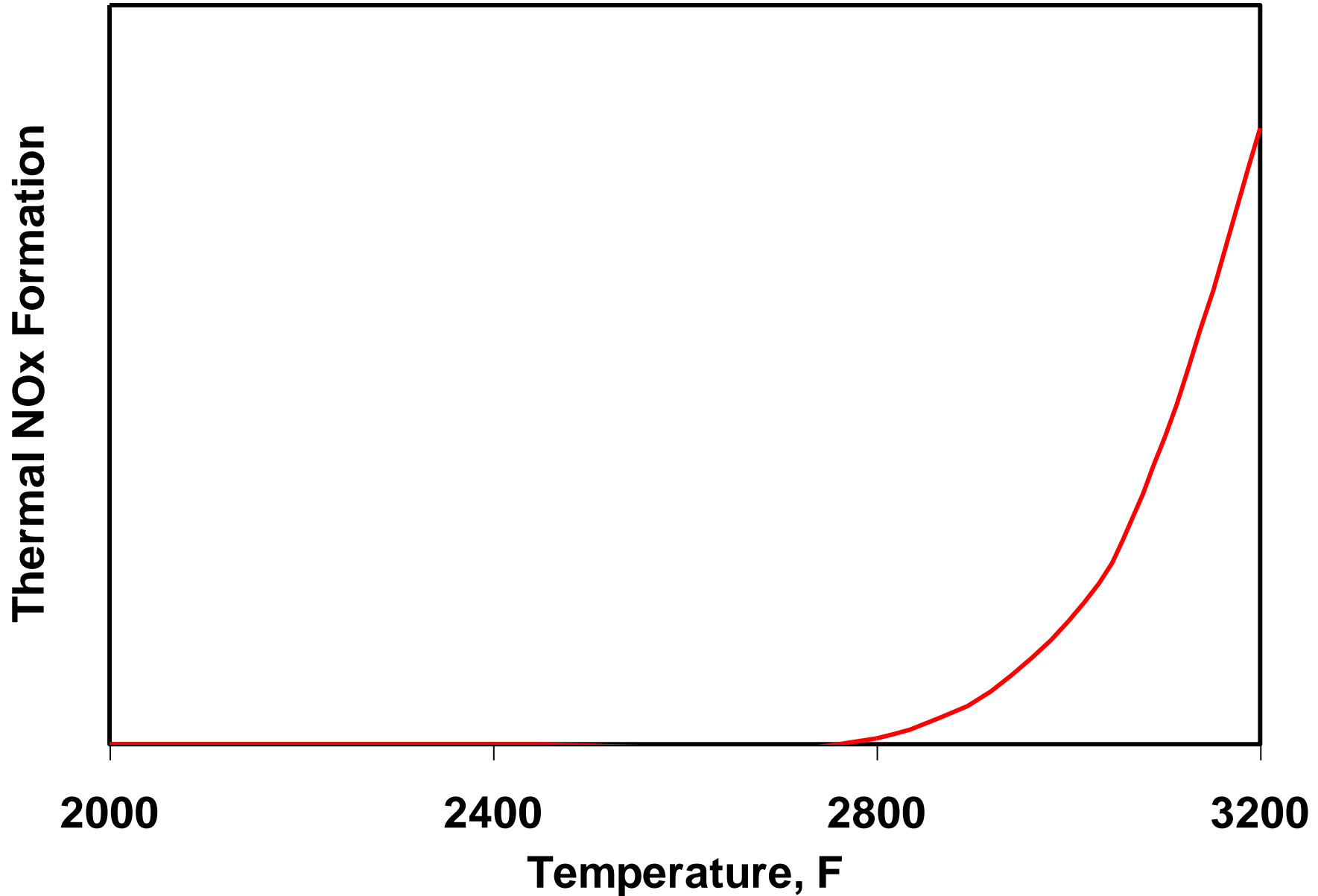


- ◆ Thermal NO<sub>x</sub>
- ◆ Fuel-bound NO<sub>x</sub>
- ◆ Prompt NO<sub>x</sub>

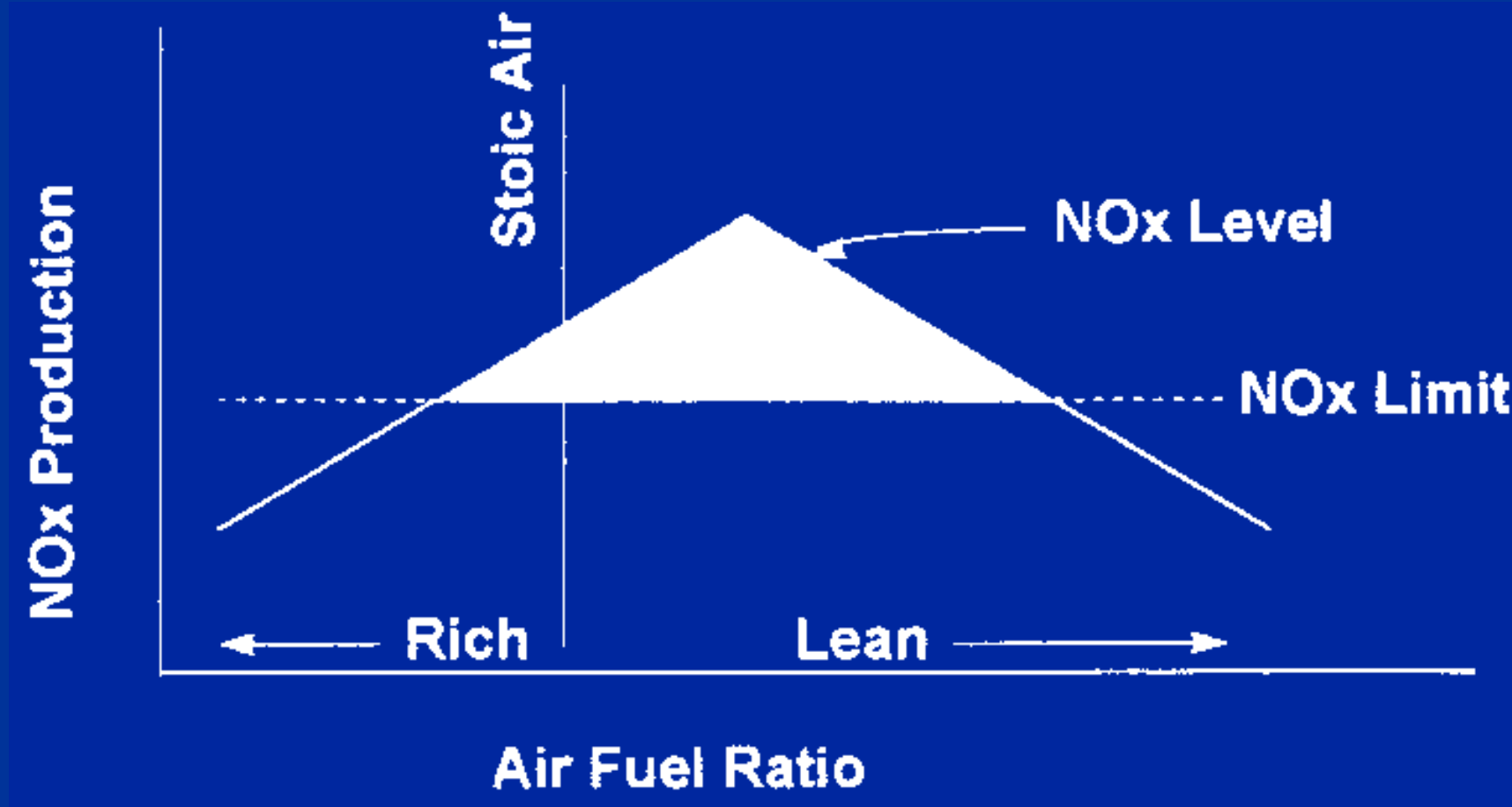
**NO<sub>x</sub> Creation**



# Thermal NOx vs. Temperature



# NOx Production vs. Air/Fuel Ratio

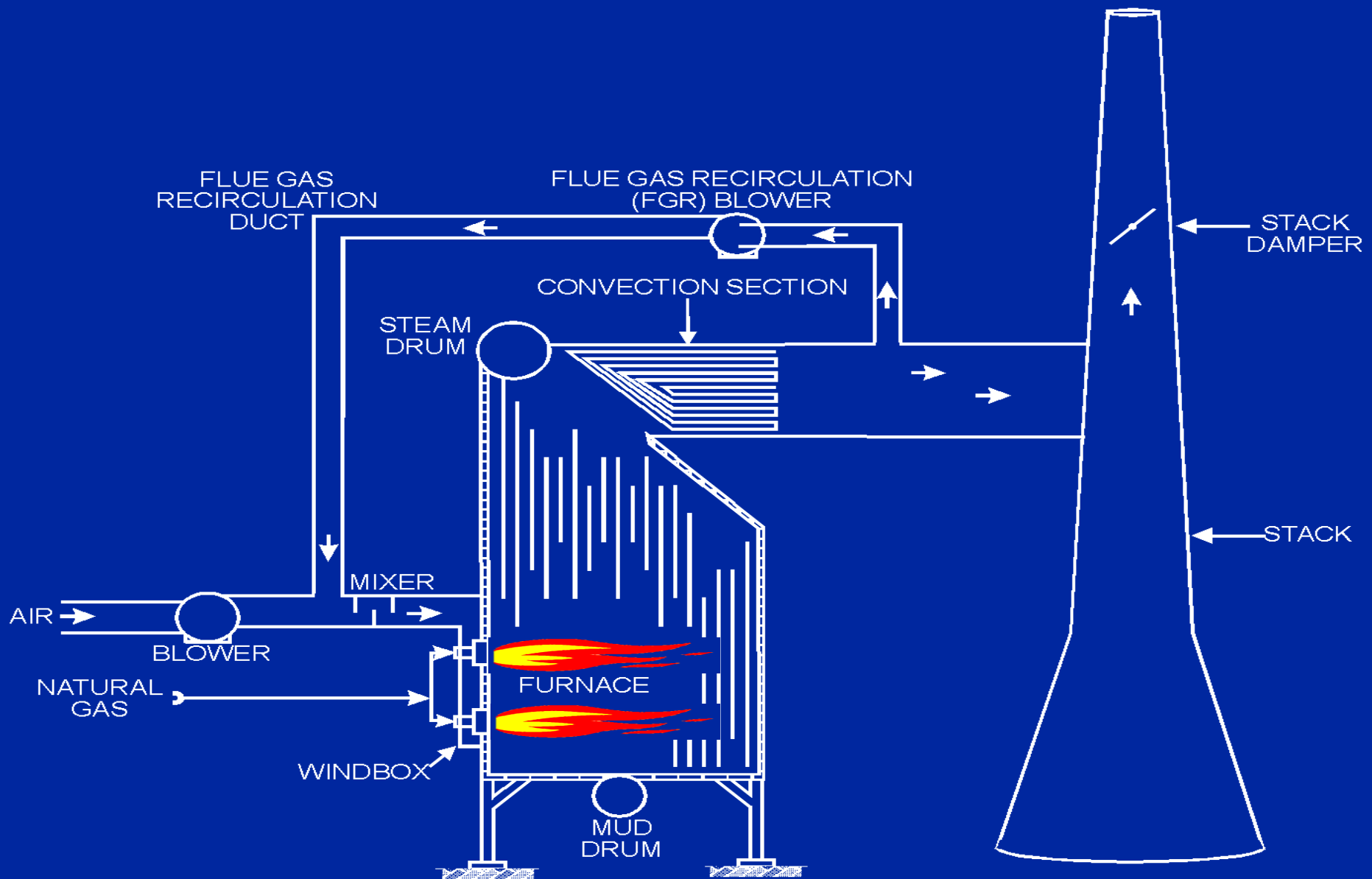




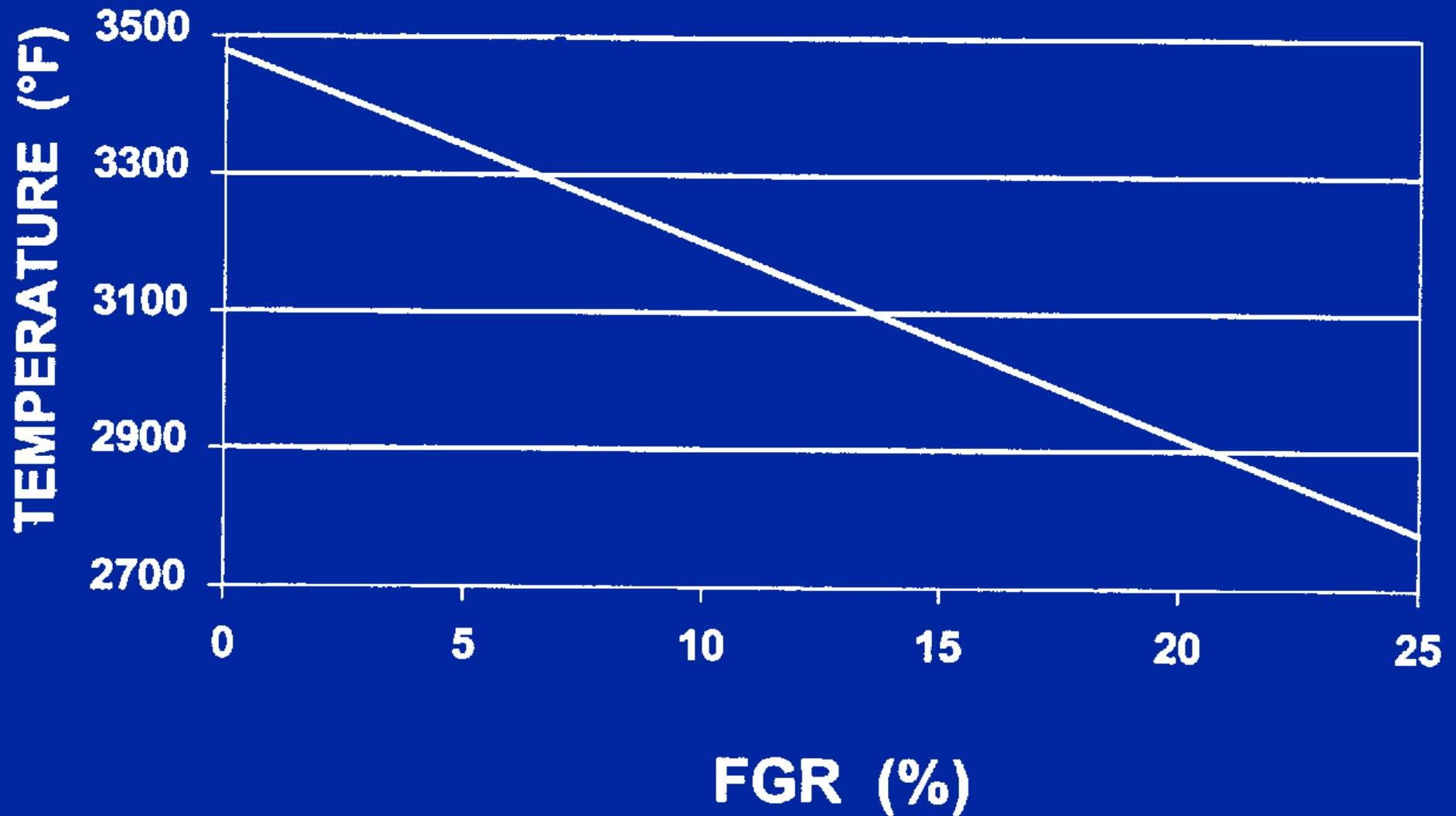
# Let's Discuss FGR



# Flue Gas Recirculation (FGR)



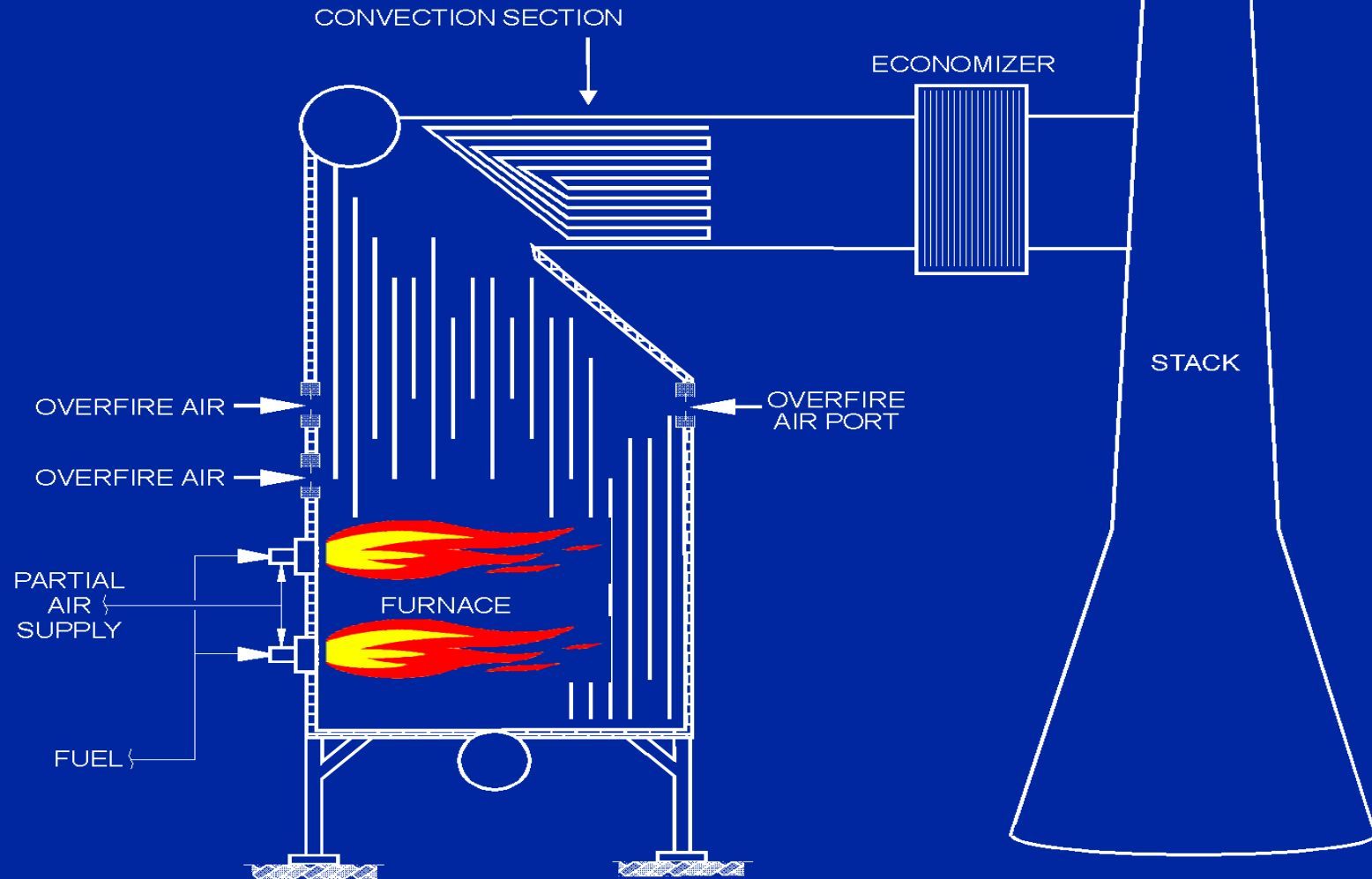
# Flame Temperature vs. FGR



A winter scene in a forest. Two large, thick tree trunks with reddish-brown bark are prominent in the foreground. The rest of the forest is filled with evergreen trees heavily laden with snow. The ground is also covered in a layer of snow. In the background, a wooden fence is visible through the trees.

**Let's Discuss  
Staged Combustion**

# Staged Combustion with Overfire Air



# Let's Discuss SCR, SNCR and RSCR



# *What is SCR?*

SCR



```
graph TD; SCR[SCR] --> S[Selective]; SCR --> C[Catalytic]; SCR --> R[Reduction];
```

**S**elective **C**atalytic **R**eduction

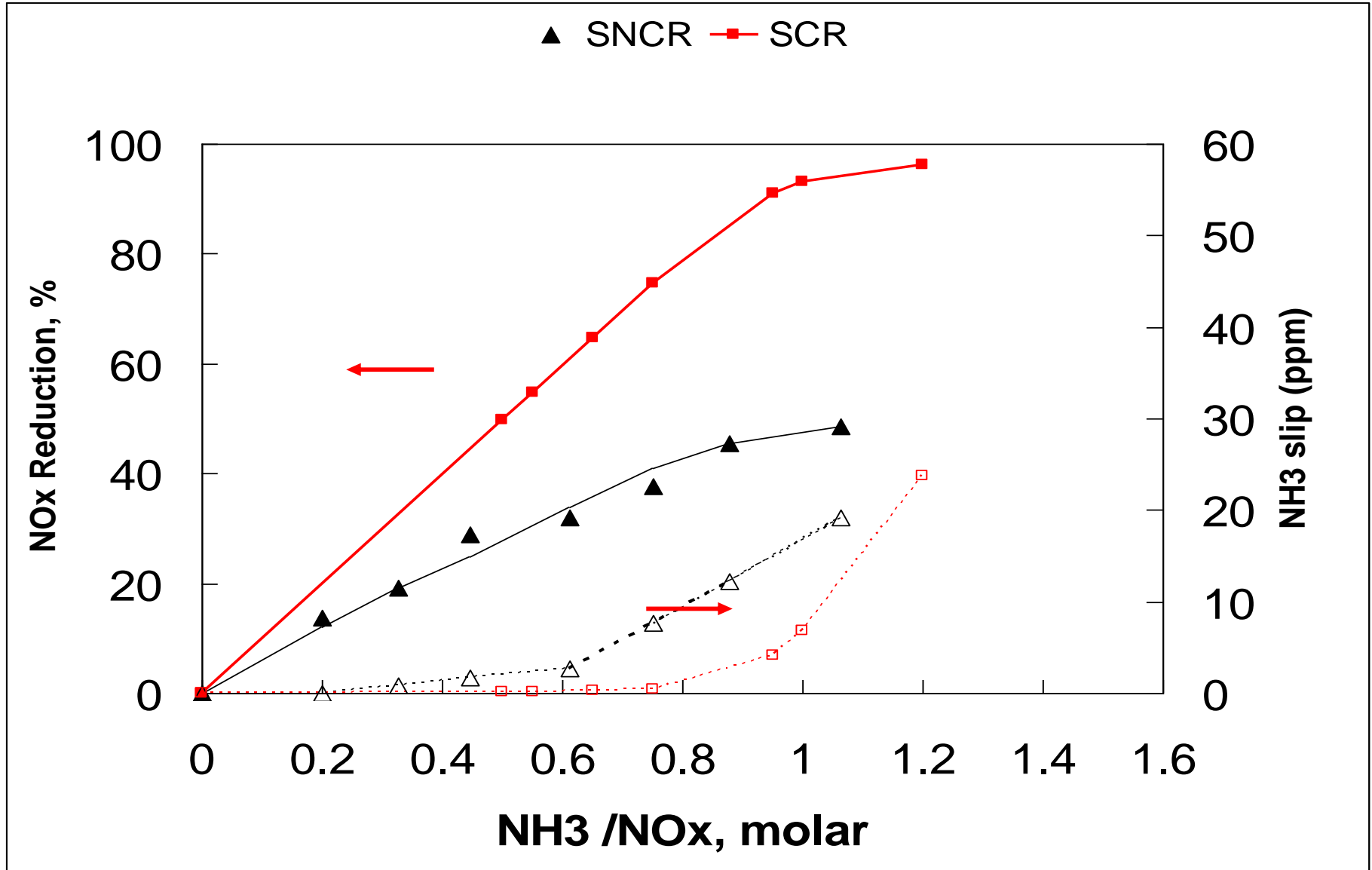
( This means that  $\text{NO}_x$  will selectively react with  $\text{NH}_3$  in the presence of Oxygen, similar to SNCR but a catalyst is needed to help the reaction which takes place at a lower temperature than SNCR)

**SCR**  
***Application***





# SNCR vs. SCR



# *Where is SCR Used*

- **Widespread Use**
  - Coal and Gas Fired Utility Boilers
  - Gas Turbine Electric Generators (Simple and Combined Cycle)
- **More Recently**
  - Refinery Combustion Systems
  - Smaller Industrial Boilers (Gas, Biomass Fired)

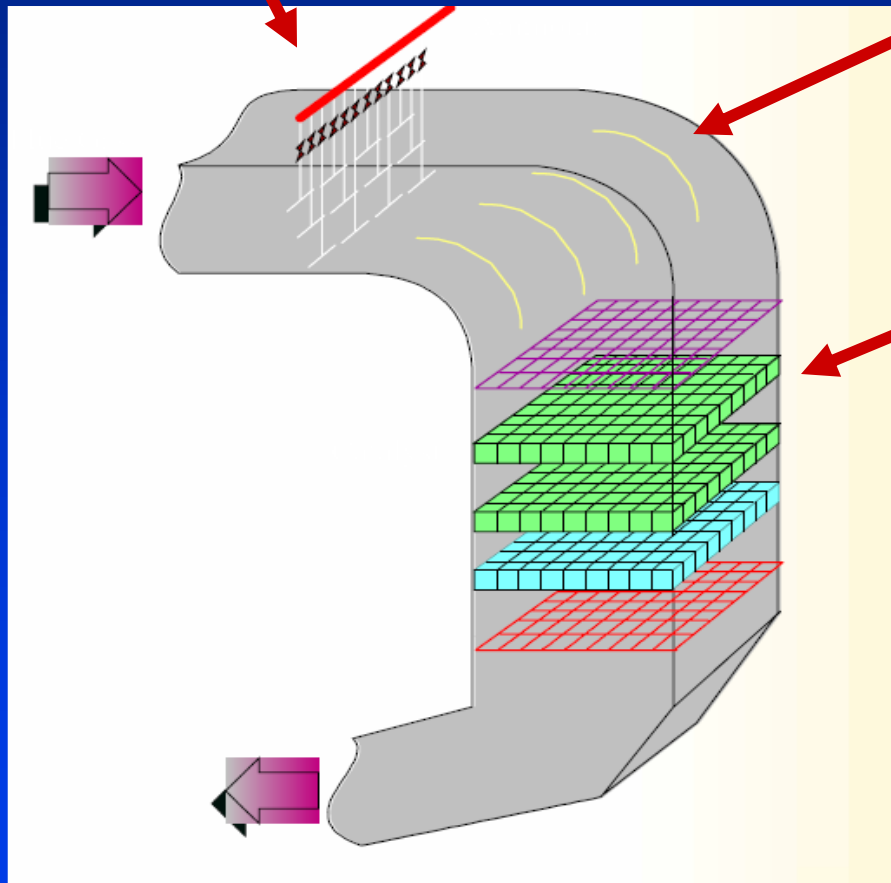
# SCR @ Typical Utility Boiler



# SCR

NH<sub>3</sub> Injection:  
(Uniform NH<sub>3</sub>/NO<sub>x</sub> Critical)

Turning Vanes to give  
uniform Velocity across the  
Catalyst

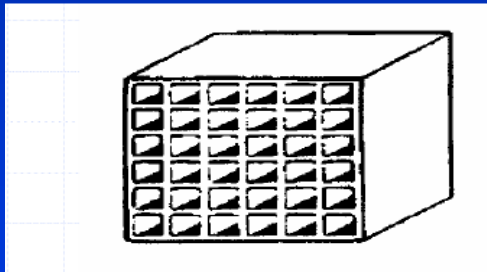
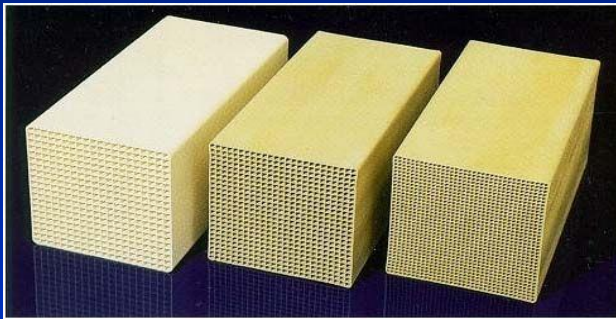


Catalyst  
Layer(s)

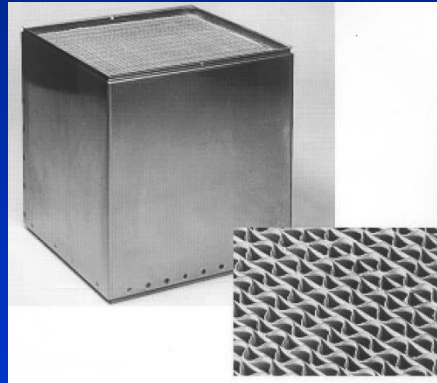


# SCR Catalyst Types

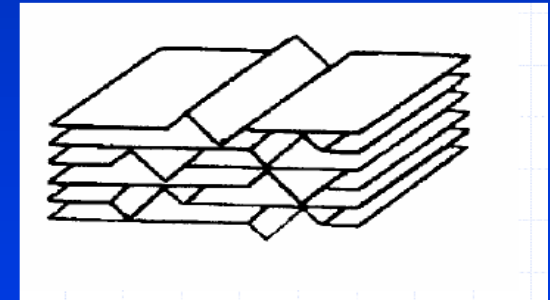
Extruded Ceramic  
Honeycomb



Corrugated  
(Haldor-Topsoe)




Plate



## Composition

- Vanadium Pentoxide ( $V_2O_5$ )
- Titanium Dioxide ( $TiO_2$ )
- Molybdenum
- Tungsten



**SCR Catalyst &  
NH<sub>3</sub> Tubes**

# SCR Catalyst

The image shows a large-scale industrial structure, possibly a Selective Catalytic Reduction (SCR) catalyst. The structure is composed of a grid of dark metal bars, creating a series of rectangular compartments. A central vertical section contains a staircase with several steps. The overall appearance is that of a complex, multi-level industrial facility. A prominent yellow text box with a red border is positioned in the upper right quadrant, containing the text "SCR Catalyst" in a bold, red, sans-serif font.

# *Catalyst Degrades With Time*

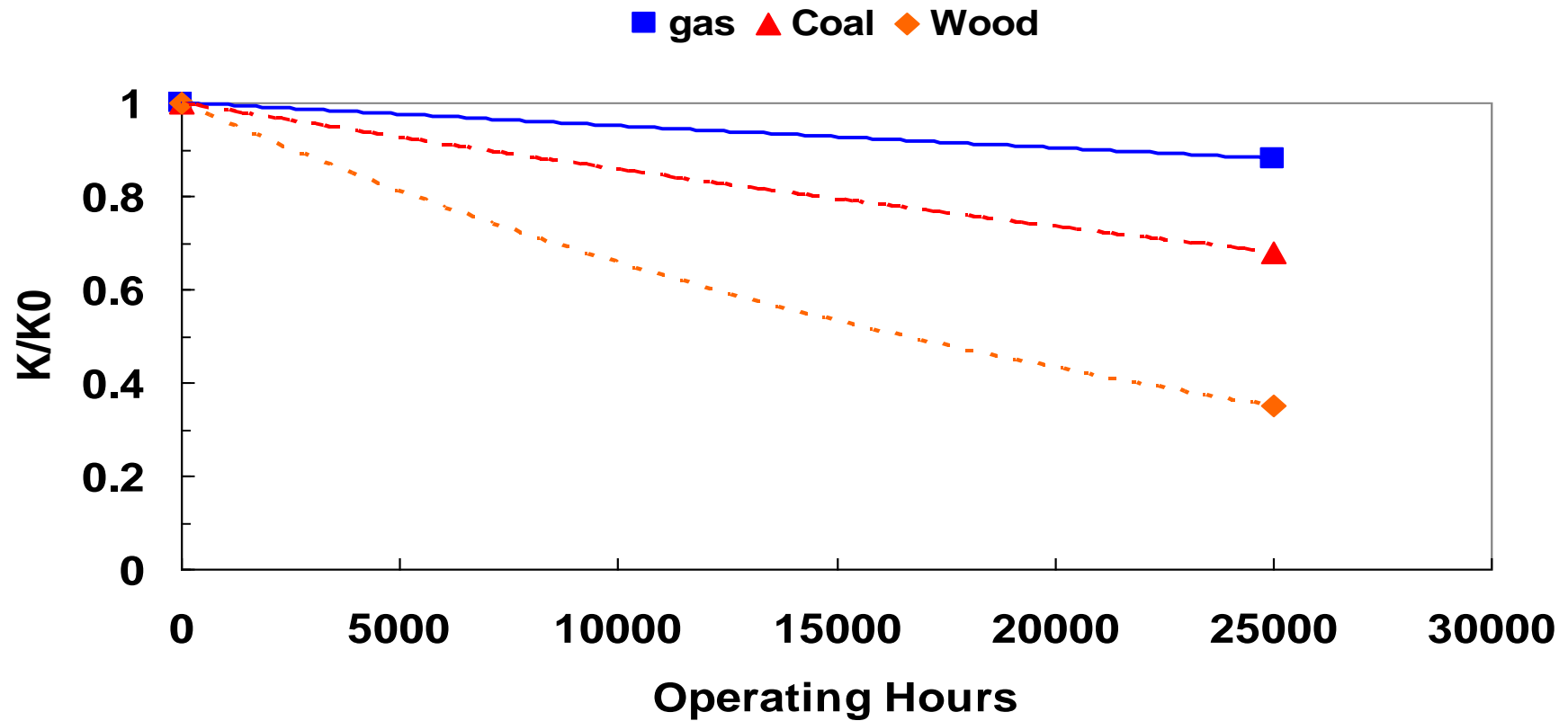
## Reason for Degradation Fuel Dependent

- ☛ Bituminous Coal-Arsenic Poisoning
- ☛ Other Coal- Calcium sulfate blinding
- ☛ Potassium & Chlorine Poisoning





# Typical Catalyst Deactivation Rates

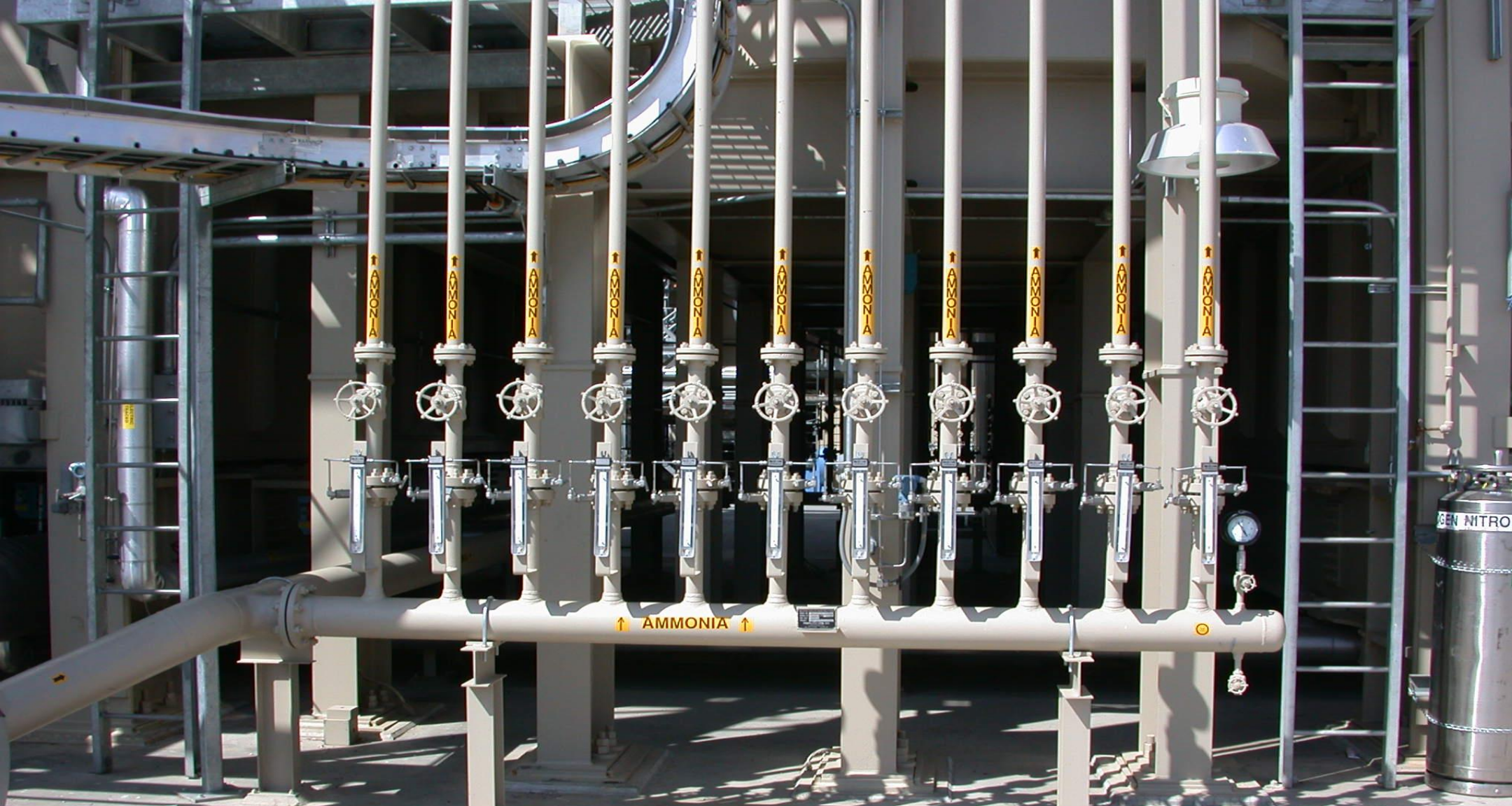


# ***Ammonia Injection Grid***





**SCR &  
NH<sub>3</sub> Tubes**



**NH<sub>3</sub> Lines**



***Small Boiler  
with SCR***



**Utility Boiler with SCR**

# ***SNCR vs SCR***

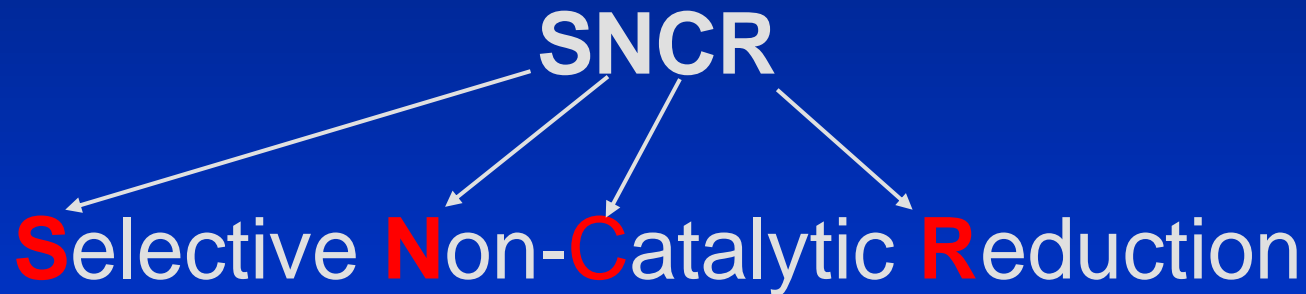
	<b>SNCR</b>	<b>SCR</b>
<b>NOx Reductiuon</b>	<b>20-50%</b>	<b>50-95%</b>
<b>Hardware</b>	<b>Simple</b>	<b>More Complex</b>
<b>Capital Cost</b>	<b>Low (1)</b>	<b>High (5-10)</b>
<b>Reagent Utilization</b>	<b>Typ. 30%</b>	<b>Almost 100%</b>
<b>O&amp;M</b>	<b>Reagent</b>	<b>Reagent/Catalyst</b>
<b>Designability</b>	<b>Poor</b>	<b>Good</b>
<b>NH3 slip</b>	<b>5-20 ppm</b>	<b>&lt;10 ppm</b>



**Let's Discuss Particulate & NH<sub>3</sub>  
Controls**



# *What is SNCR?*



(Means that a chemical will selectively react with  $\text{NO}_x$  in the presence of Oxygen)

• Ammonia ( $\text{NH}_3$ )

Urea ( $\text{NH}_2\text{CONH}_2$ )



**Circulating Fluidized Bed Boiler  
w/SNCR**

# ***SNCR @ Bio-Mass Plant***



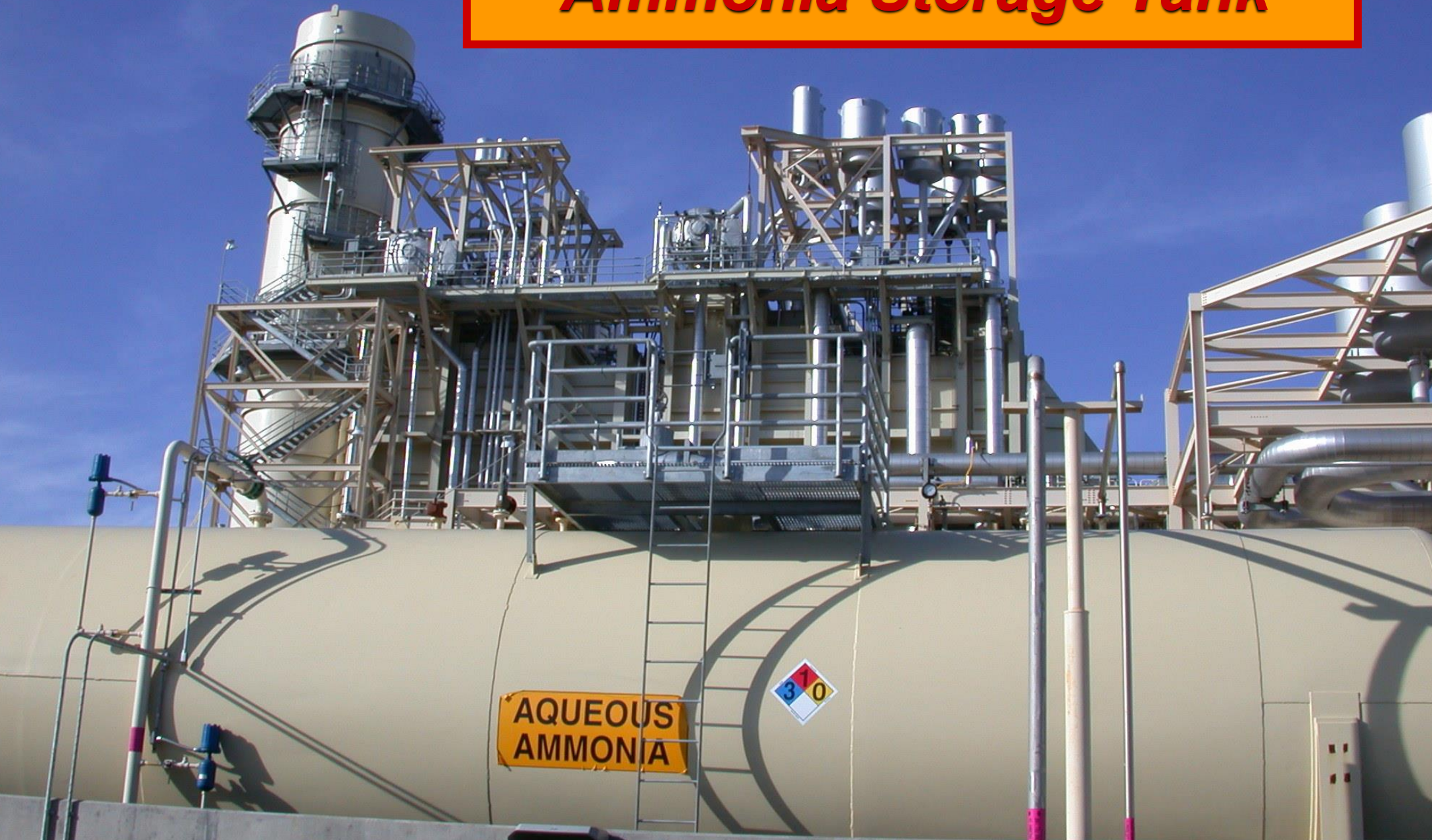


**Superheater**

**SNCR**

**Bio-Mass Boiler**

# *Ammonia Storage Tank*



# Anhydrous Ammonia Storage Tank



# ***Ammonia vs. Urea***

<b>Parameter</b>	<b>Ammonia</b>	<b>Urea</b>
<b>Form</b>	<b>High Vapor Pressure Liquid Ammonia/Water Solution</b>	<b>Liquid Solution</b>
<b>Safety</b>	<b>Anhydrous/29.4% Aqueous- safety iss 19% Aqueous- fewer Safety Issues</b>	<b>No Safety Issues</b>
<b>Storage</b>	<b>Anhydrous-Pressure Vessel Aqueous- Atmospheric Pressure</b>	<b>Atmospheric Pressure Crystalization at Low Temperature</b>
<b>Injectors</b>	<b>Needs Carrier Gas</b>	<b>Atomizers( Pressure or Twin Fluid)</b>
<b>Temperature</b>	<b>Peak Removal @ 1750 F</b>	<b>Peak Removal @ 1850 F Large Dilute Drops Shield Urea</b>
<b>System Complexity</b>	<b>Relatively Simple</b>	<b>Relatively Simple</b>



**Bio-Mass Boiler**



**Disconnected  
NH<sub>3</sub> Line**



# ***Balance-of-Plant Impacts***

- ◆ **NH<sub>3</sub> Slip**
- ◆ **SO<sub>3</sub>/NH<sub>3</sub> Reactions (APH Deposition)**
- ◆ **HCl/NH<sub>3</sub> Reactions (Plume Visibility)**
- ◆ **Ash/NH<sub>3</sub> Absorption (Ash Sales, General Nuisance)**
- ◆ **N<sub>2</sub>O Emissions**

# *What is RSCR?*

RSCR

```
graph TD; RSCR --> Regenerative; RSCR --> Selective; RSCR --> Catalytic; RSCR --> Reduction;
```

**Regenerative Selective Catalytic Reduction**

( This means that  $\text{NO}_x$  will selectively react with  $\text{NH}_3$  in the presence of Oxygen, similar to SCR with a catalyst to help the reaction and two thermal transfer beds)

## *Ammonia Slip*

- ◆  $\text{NH}_3 + \text{OH} \Rightarrow \text{NH}_2 + \text{H}_2\text{O}$
- ◆  $\text{NH}_2 + \text{NO} \Rightarrow \text{N}_2 + \text{H}_2\text{O}$
- ◆  $2\text{NH}_3 + \text{OH} + \text{NO} \Rightarrow 2\text{H}_2\text{O} + \text{N}_2 + \text{NH}_3$
- ◆ 10 to 25 ppm  $\text{NH}_3$  Slip
- ◆ Could be higher
- ◆ Always have Some  $\text{NH}_3$  slip

# $\text{NH}_4\text{Cl}$ Formation



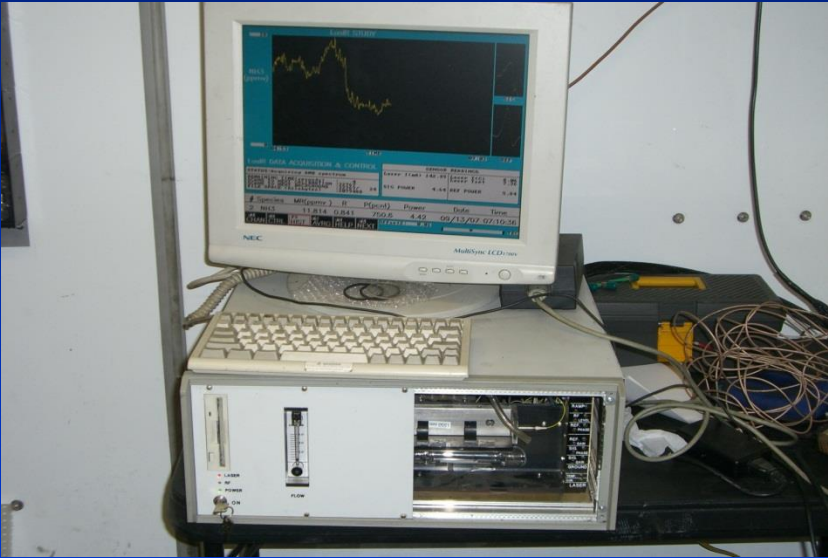
## ***NH<sub>4</sub>Cl Formation***

- ◆ **Function of the concentrations of NH<sub>3</sub> and HCl**
- ◆ **Concentrations decrease as air is mixed into the plume**
- ◆ **Lower concentrations => less NH<sub>4</sub>Cl formed**
- ◆ **Therefore: air dilution is good**

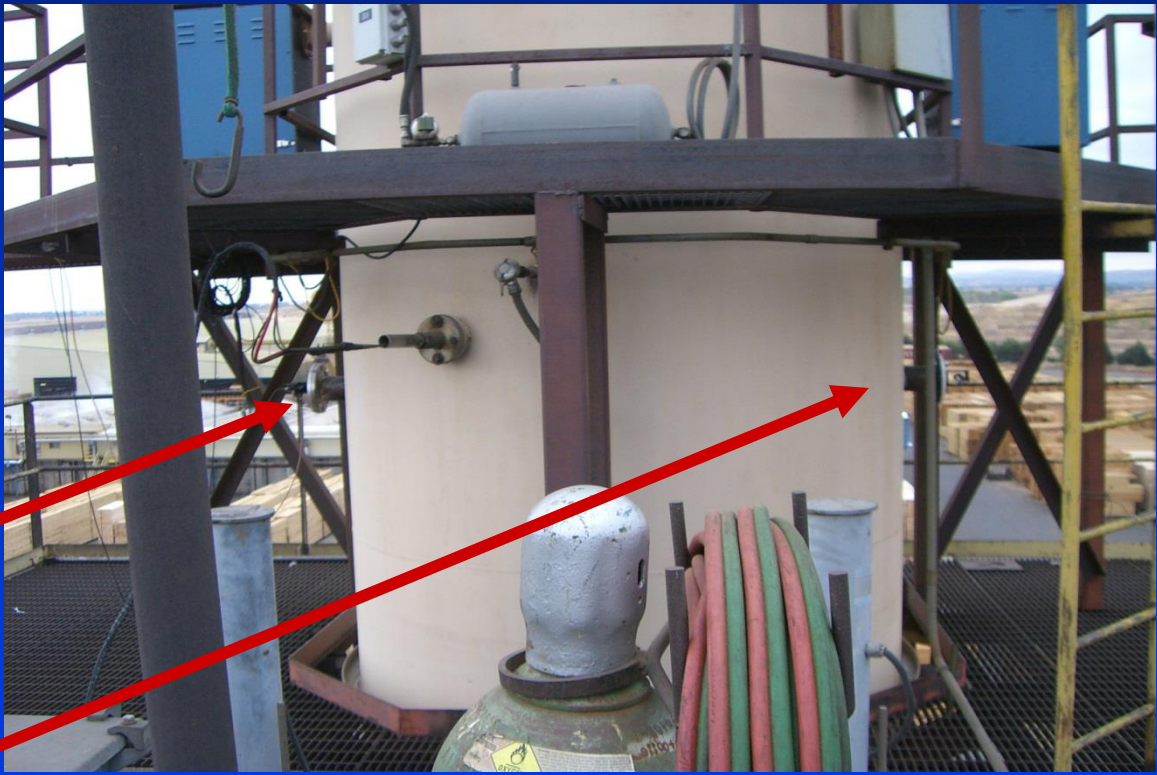
# ***What Can Be Done??***

- ◆ **Minimize (eliminate Cl) in fuel**
- ◆ **Install acid gas controls**
- ◆ **Minimize NH<sub>3</sub> slip <= monitor**
- ◆ **High stack gas temperatures**
- ◆ **High ambient air temperatures (winter time a problem??)**
- ◆ **Promote rapid gas/air mixing ??**
- ◆ **Install high gas temperature concentric stack annulus ??**

# Continuous NH3 Analyzer



Laser & Detector



Retro Reflector



# Comparison of NOx Reduction Technologies

Technology	Approx. Reduction	Approx. lbs/MMBTU	Approx. ppmv @ 3% O2
Standard burners	Base case	0.14	120
Low NOx burners	60%	0.06	45
Ultra Low NOx burners – 1 <sup>st</sup> gen.	80%	0.03	25 - 30
Ultra Low NOx burners – 2 <sup>nd</sup> gen.	95%	0.007	6 - 9
FGR	55%	0.025	20
Compu- NOx w/ FGR	90%	0.015	15 - 20
SNCR	80%	0.033 - 0.085	27 - 70
Catalytic Scrubbing	70%	0.017 - 0.044	14 - 36
SCR	90 – 95%	0.006 - 0.015	5 - 12

# Let's Discuss PM Control



# ***What is Particulate Matter??***

- ◆ **It is what the test measurement says it is**
- ◆ **Meaning:**
  - ◆ **Solid particles that are captured on a filter**
  - ◆ **Condensable matter collected in a set of impingers**
- ◆ **What eventually condenses in the atmosphere is also considered as particulate matter along with “solid” particulate in the gas stream**



**PM**



**Circulating  
Fluidized Bed  
Boiler**

***Sources of PM***

## **Sources of “Particulate Matter”**

- ◆ **Ash in the fuel**
  - ◆ **Silica and Alumina - generally large particles that are retained or collected in the boiler/precipitator**
  - ◆ **Intrinsic ash - generates the small particles that are more troublesome to control**
  - ◆ **Alkalis - potassium, sodium and calcium**
- ◆ **Condensables (HCl, SO<sub>3</sub>, NH<sub>4</sub>Cl) which are also considered as “particulates**

# Control of Particulate Emissions

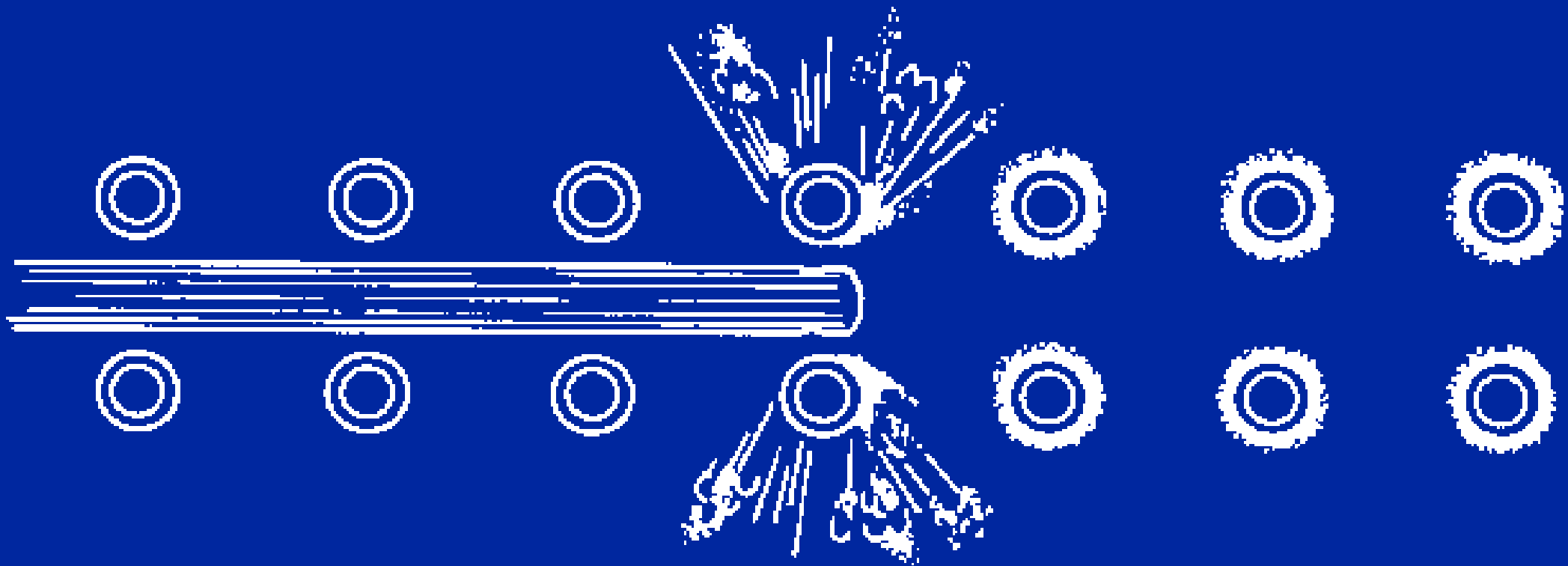
- ◆ Settling chambers
- ◆ Cyclones
- ◆ Baghouses
- ◆ ESPs
- ◆ Scrubbers



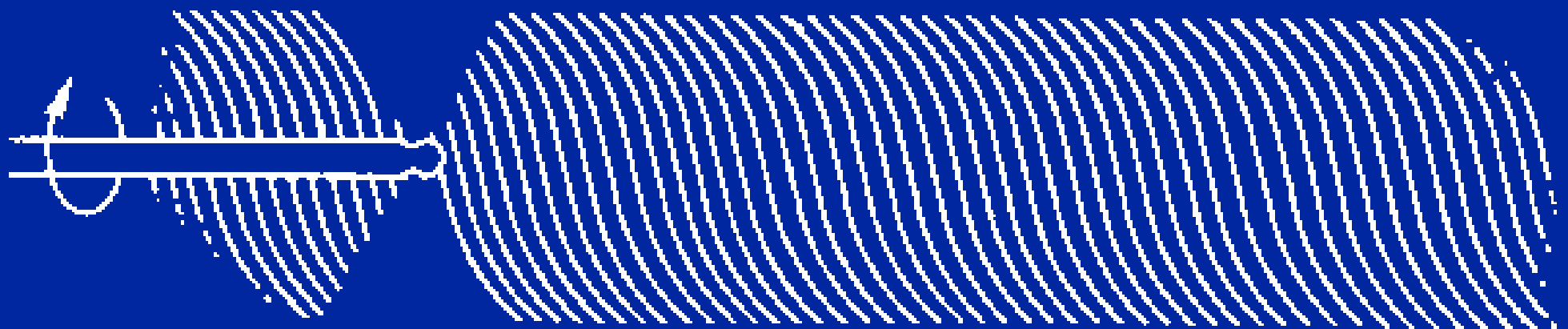
# Water Spray



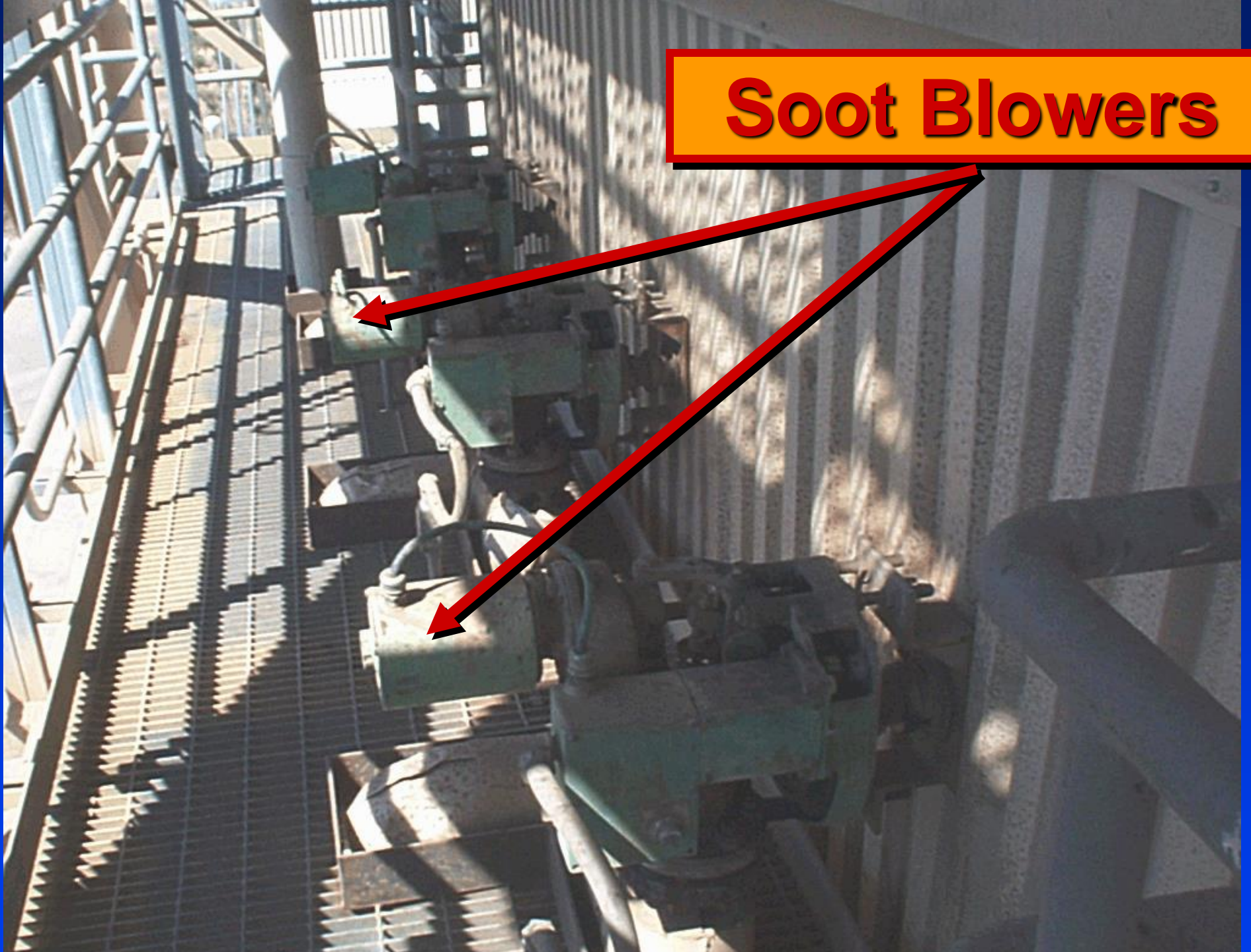




# Soot Blowing



# Soot Blowers





**Cyclone**



# Multi- Cyclone



**Baghouse**

**ESP**





# Regulatory Requirements

# Regulatory Requirements

- ◆ **Federal, state, and local requirements**
- ◆ **Boiler specific limits**
- ◆ **Permit requirements**
- ◆ **Monitoring requirements**
- ◆ **Visible emission limits**
- ◆ **Nuisance regulations**
- ◆ **Breakdowns & variances**





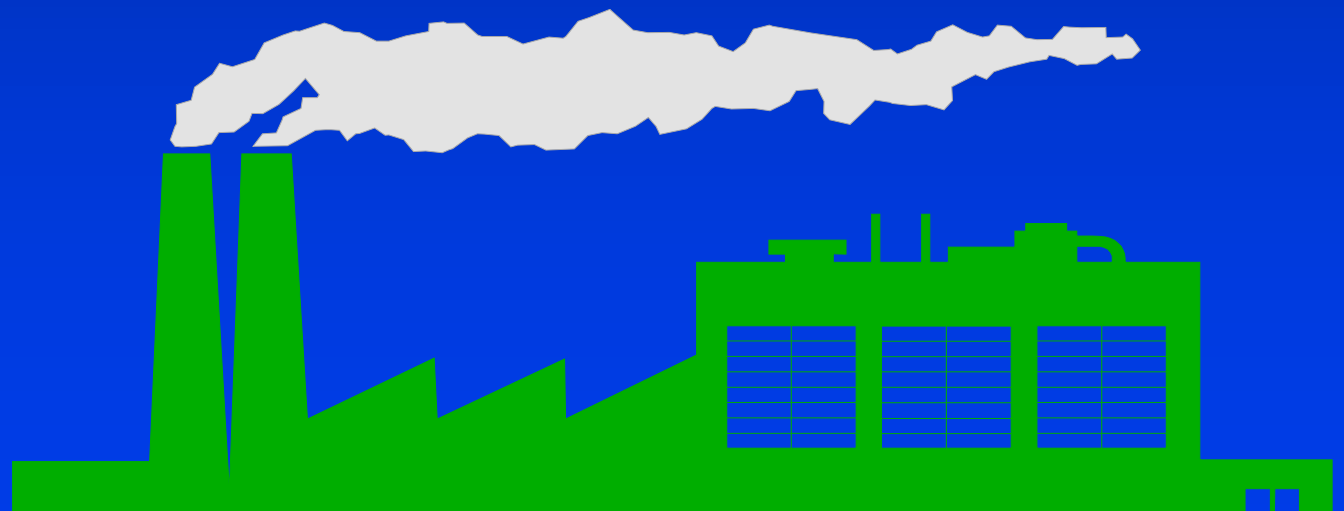
# Boiler Regulations

- ◆ NSPS 40 CFR Part 60 Subpart D, Da, Db, Dc, Ea
- ◆ Acid Rain Provisions (Parts 72,73,74,75, 76, 77, 78)
- ◆ RCRA 40 CFR Parts 264 & 266
- ◆ State Regulations including VE
- ◆ SIP Requirements
- ◆ Local Regulations
- ◆ MACT DDDDD & JJJJJJ



# Boiler Emission Limits

- ◆ **NO<sub>x</sub>, SO<sub>2</sub>, particulate, and opacity values for boilers are based on applicable subpart, heat input, date built or modified, and fuel used**
- ◆ **States and districts may have more stringent limits**

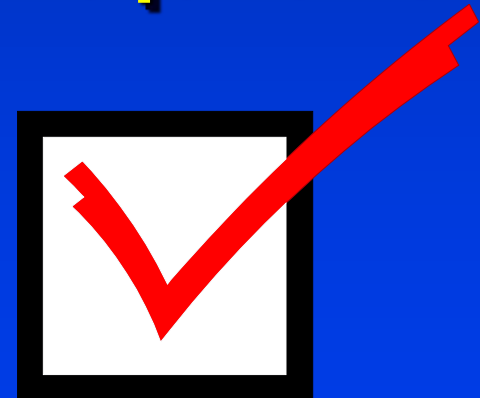


# BACT

Type of Control	NOx Limits
Natural Gas Fired with SCR & Low NOx Burner	0.010 lb/MMBTU
Natural Gas Fired Units (< 60 MMBTU/hr)	0.035 lb/MMBTU
Biomass Fuel Fired Boilers (Large), SNCR	0.10 lb/MMBTU
Municipal Solid Waste	110 ppmv @7% O2

# Permit Condition Categories

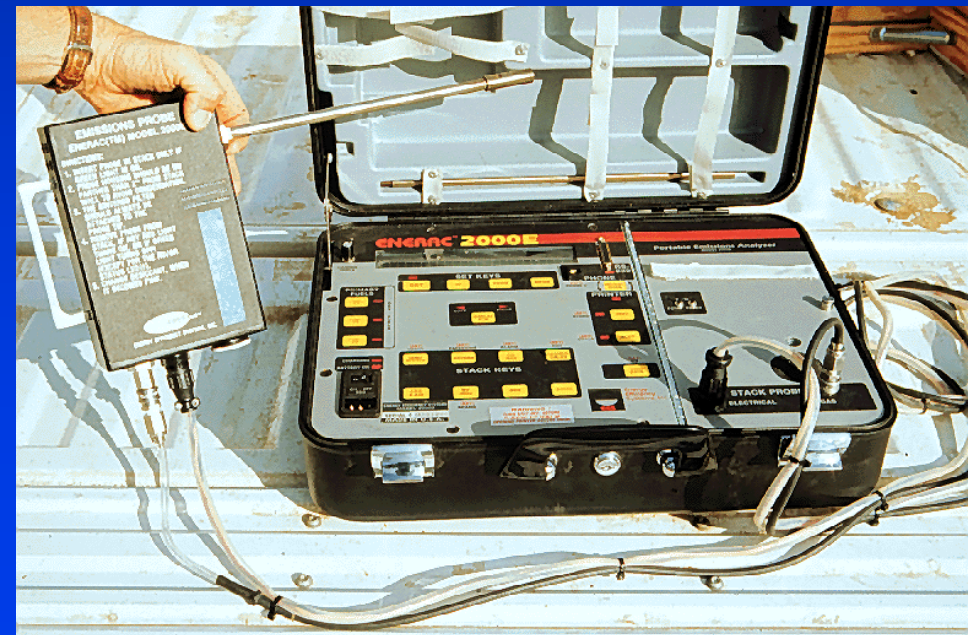
1. Emissions Limitations
2. Equipment Requirements
3. Operating Conditions
4. Monitoring and Recording Requirements
5. Compliance Testing
6. General Requirements



# Alternative Monitoring

- ◆ Portable analyzer monitoring of NO<sub>x</sub>, CO, O<sub>2</sub>
- ◆ Determination of FGR rate
- ◆ Burner mechanical adjustments
- ◆ O<sub>2</sub> Trim concentration
- ◆ FGR valve(s) setting

**Portable Combustion  
Analyzer**





# Boiler Inspections

# Points of Inspection

- ◆ Capture
- ◆ Transport
- ◆ Air mover
- ◆ Control device
- ◆ Instrumentation
- ◆ Subsystem
- ◆ Records



# Pre-Inspection

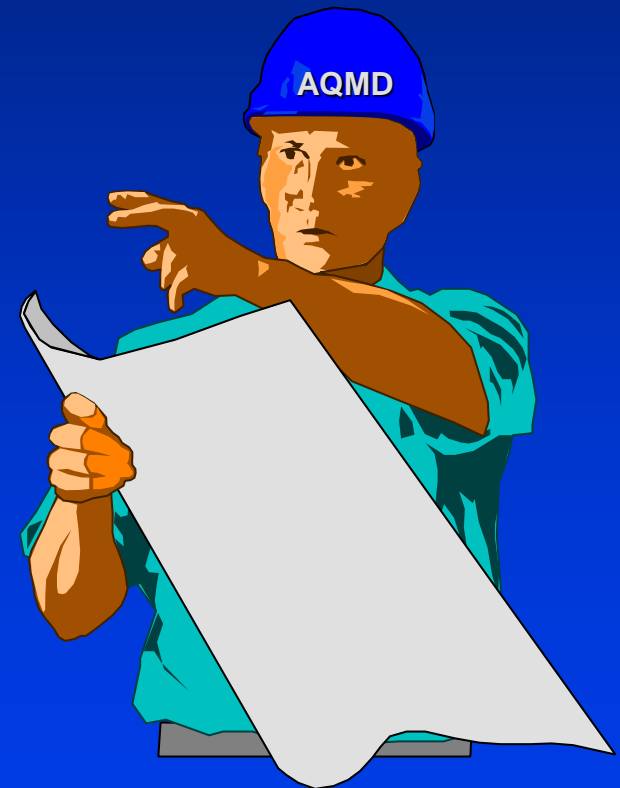


- ◆ Prepare inspection form
- ◆ File review
- ◆ Regulation review
- ◆ Equipment check
- ◆ Pre-entry & entry
- ◆ Pre-inspection meeting
- ◆ Permit check



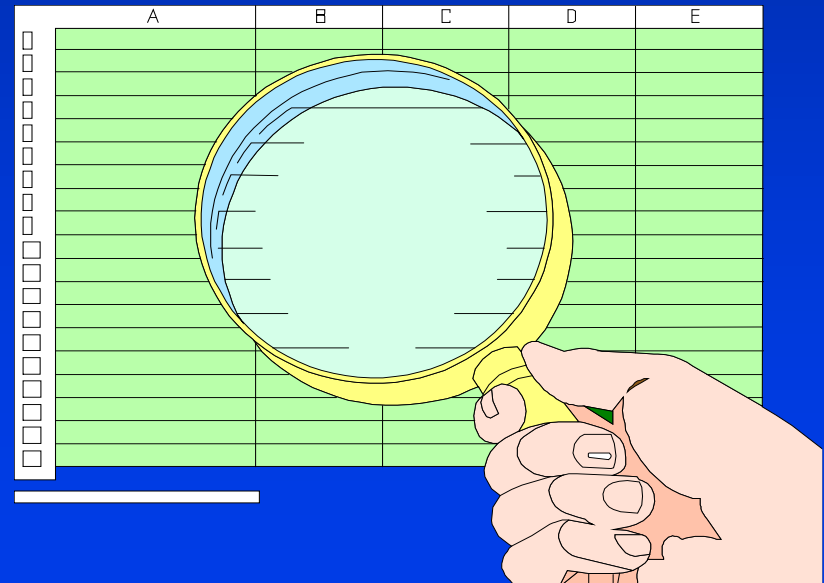
# Reasons for Inspections

- ◆ **Compliance determination**
- ◆ **Complaint investigation**
- ◆ **Source plan approval**
- ◆ **Review or renewal of permits**
- ◆ **Special studies**



# Inspection

- ◆ **Visible emission evaluation**
- ◆ **General upkeep & maintenance**
- ◆ **Monitoring instruments & records**
- ◆ **Fuel type and quality**
- ◆ **Maintenance records**
- ◆ **Operational records**
- ◆ **Source tests**

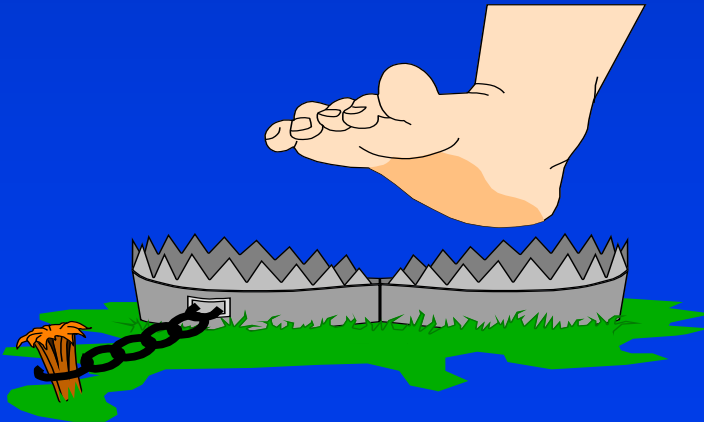


# Plant Safety



# Inspector Safety

- ◆ Proper equipment
- ◆ Plant warnings
- ◆ Heat
- ◆ High pressure steam
- ◆ Electrical hazards
- ◆ Noise
- ◆ Moving parts
- ◆ Inhalation hazards
- ◆ Hazardous materials
- ◆ Machine disintegration
- ◆ Fires
- ◆ Other hazards & traps





**Plant Safety**



**Plant Hazards**

**Confined  
Space**



**DANGER**  
**CONFINED SPACE**  
ENTER BY PERMIT ONLY

**Confined Space**

**DANGER**  
**CONFINED SPACE**  
POSSIBLE NITROGEN  
ATMOSPHERE  
ADEQUATE MECHANICAL  
VENTILATION REQUIRED



**DANGER**

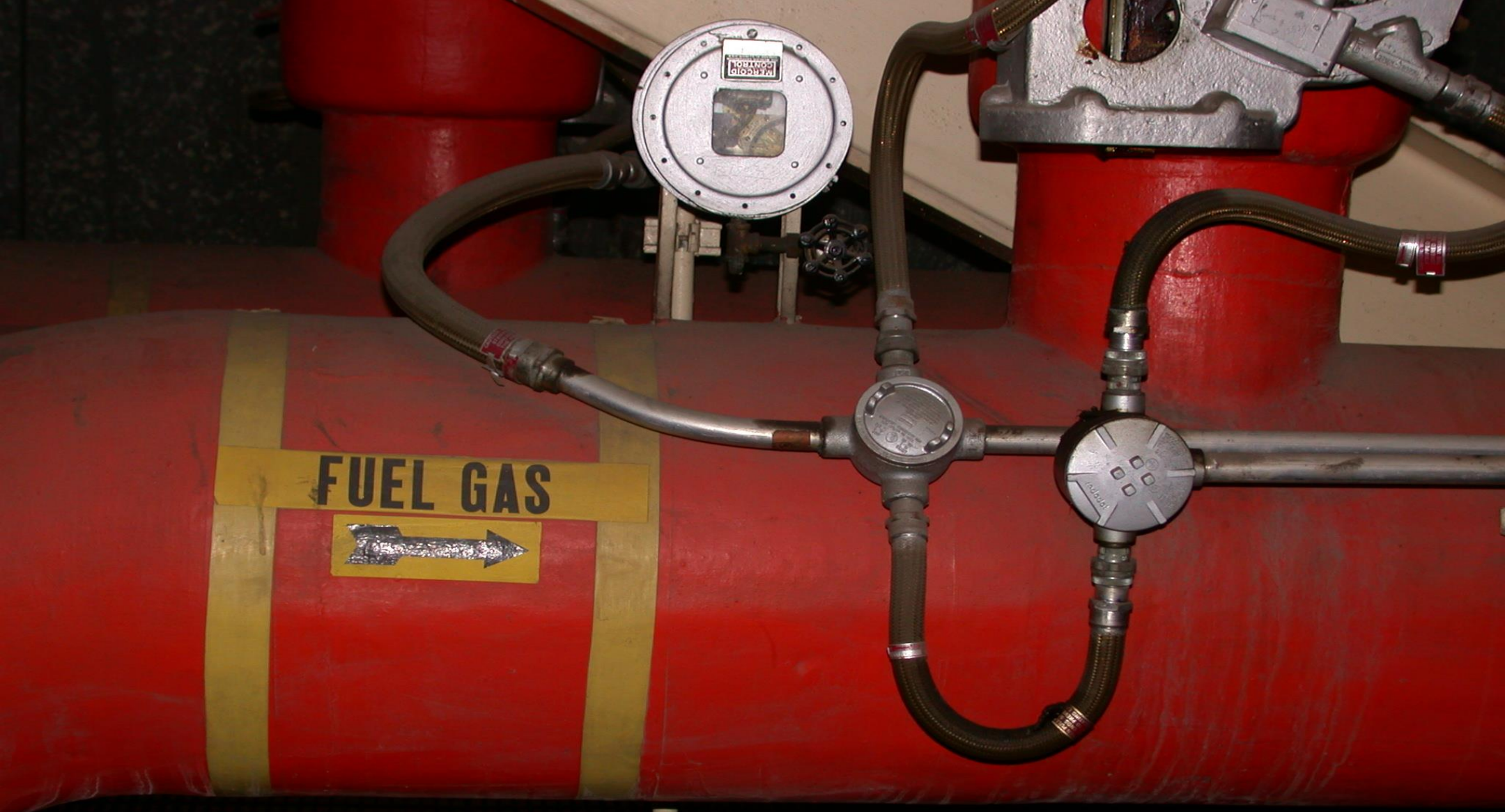
**CONTAINS ASBESTOS FIBERS**

**AVOID CREATING DUST**

**CANCER AND LUNG DISEASE HAZARD**

**AVOID BREATHING AIRBORNE ASBESTOS**


62-1397 (5/88)



**HP Gas Lines**

A photograph showing a significant structural failure in a wall. A large, jagged hole has been made in the drywall, exposing a network of pipes and conduits. The scene is illuminated by a bright, warm light source, likely a flashlight, which creates a strong glare and casts shadows. The exposed pipes appear to be made of metal and are surrounded by insulation and other building materials. The overall appearance is one of a major industrial or utility accident.

**Ruptured  
Steam Line**

A photograph of an industrial steam exhaust system. In the center, a large, cylindrical motor is mounted on a concrete base, emitting a thick plume of white steam. The motor is surrounded by various pipes and machinery. In the foreground, there are large, vertical, cylindrical components wrapped in grey insulation. The background shows a dimly lit industrial space with more equipment and a red emergency stop button on a wall. The overall scene is characterized by a warm, yellowish light and a hazy atmosphere due to the steam.

**Steam  
Exhaust**

**Access**



**Access**



**Thank You!**

